

Periodic Table and Periodic Law

1. On the blank periodic table, *clearly* indicate six of the following element sections, using your own color code.

a. Representative elements	h. Alkali metals
b. Transition elements	i. Alkaline earth metals
c. Inner-transition elements	j. Halogens
d. Chalcogens	k. Noble gases
e. Coinage metals	l. Noble metals
f. Metalloids	m. Lanthanide series
g. Post-transition metals	n. Actinide series

A blank periodic table grid. The main body consists of 7 rows and 18 columns. The first two columns are on the left, and the last two columns are on the right, with a gap in the middle. Below the main body, there are two separate rows of 14 cells each, representing the lanthanide and actinide series.

- Sketch in the staircase line that separates the metals from the nonmetals on the periodic table.
- Identify the atomic numbers of the elements that would be called the *transactinide* elements.

Instructors Manual For Periodic Law Experiment

Tobias Bleicker



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