

# P6 - REVISION

- 1) This paper is an alternative to a practical exam, so each experiment is exactly as described or drawn.
- 2) Students should understand why the choice of range for the measuring scale should match the size of the quantity being measured.
- 3) Students should know how to record measurements in a table. A table should record all the measurements needed to obtain the value of a given physical quantity. Columns (or rows) in the table should be headed with the symbol of the physical quantity and its unit (e.g.  $T^2/s^2$ ). The SI units are recommended. Neat work is required. If an equation is used to calculate a quantity it must be mentioned.

For example,

L/cm	T/s	$T^2/s^2$	R/ $\Omega$
$\theta / ^\circ$	$\theta / ^\circ\text{C}$	V/V	I/A
m/g	V/cm <sup>3</sup>	$\frac{1}{T^2} / \frac{1}{s^2}$	$\frac{m}{L} / \frac{g}{cm}$

- 4) Ideally, when performing an experiment (and relevant readings are recorded) it is helpful to arrange the experiment so that one variable (the independent one) is changed step by step then the dependent variable is recorded. Candidates should always look for a trend in the recorded results.

Some trends are:

- y increases as x increases
  - Straight line through the origin,  $x$  is doubled then  $y$  is doubled, so direct proportionality ( or  $\frac{y_2 - y_1}{x_2 - x_1} = \text{constant}$  )
  - y decreases as  $x$  increases
  - If  $x$  is doubled and  $y$  is halved or  $y = k$ , so inverse proportionality.
  - **SOME RELATIONS ARE LINEAR BUT NOT DIRECT OR INVERSE.**
- 5) Students should understand the idea of a fair test or comparison in which only one variable is altered at a time, e.g. when investigating how rate of cooling experiment depends on initial temperature of the liquid then the room temperature, room draughts, volume and type of liquid, rate of stirring to be kept constant.

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