$$2C_2H_2 + 5O_2 \rightarrow 4CO_2 + 2H_2O$$

2 vol. 5 vol. 4 vol. 2 vol.

 $8.4 \, dm^3$? vol.

MeritBatch.com If 4 vol. of CO₂ \rightarrow 2 vol. of C₂H₂

then 1 vol. of
$$CO_2 \rightarrow \frac{2}{42}$$
 vol. of C_2H_2

8.4 dm³ of CO₂
$$\rightarrow \frac{1}{2} \times 8.4$$
 dm³ of C₂H₂

 $4.2 \text{ dm}^3 \text{ of C}_2H_2$

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