

ICSE 2023 EXAMINATION
SPECIMEN QUESTION PAPER
COMPUTER APPLICATIONS

Maximum Marks: 100

Time allowed: Two hours

Answers to this Paper must be written on the paper provided separately.

*You will **not** be allowed to write during the first 15 minutes.*

This time is to be spent in reading the question paper.

The time given at the head of this Paper is the time allowed for writing the answers.

*This Paper is divided into **two** Sections.*

*Attempt **all** questions from **Section A** and **any four** questions from **Section B**.*

The intended marks for questions or parts of questions are given in brackets[].

SECTION A

*(Attempt **all** questions from this **Section**.)*

Question 1

[20]

Choose the correct answer and write the correct option.

- (i) Wrapping up of data and methods together as one unit is termed as:
- (a) Inheritance
 - (b) Polymorphism
 - (c) Encapsulation
 - (d) Abstraction
- (ii) The datatype which is specified that the method does not return a value is:
- (a) Void
 - (b) void
 - (c) VOID
 - (d) boolean

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Table of Contents Icse Computer Applications Sample Papers

1. Understanding the eBook Icse Computer Applications Sample Papers
 - The Rise of Digital Reading Icse Computer Applications Sample Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Icse Computer Applications Sample Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Icse Computer Applications Sample Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Icse Computer Applications Sample Papers
 - Personalized Recommendations
 - Icse Computer Applications Sample Papers User Reviews and Ratings
 - Icse Computer Applications Sample Papers and Bestseller Lists

5. Accessing Icse Computer Applications Sample Papers Free and Paid eBooks
 - Icse Computer Applications Sample Papers Public Domain eBooks
 - Icse Computer Applications Sample Papers eBook Subscription Services
 - Icse Computer Applications Sample Papers Budget-Friendly Options
6. Navigating Icse Computer Applications Sample Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Icse Computer Applications Sample Papers Compatibility with Devices
 - Icse Computer Applications Sample Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Icse Computer Applications Sample Papers
 - Highlighting and Note-Taking Icse Computer Applications Sample Papers
 - Interactive Elements Icse Computer Applications Sample Papers
8. Staying Engaged with Icse Computer Applications Sample Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Icse Computer Applications Sample Papers
9. Balancing eBooks and Physical Books Icse Computer Applications Sample Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Icse Computer Applications Sample Papers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Icse Computer Applications Sample Papers
 - Setting Reading Goals Icse Computer Applications Sample Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Icse Computer Applications Sample Papers
 - Fact-Checking eBook Content of Icse Computer Applications Sample Papers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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