

Empirical Formulas Practice Packet

Show all Calculations. Write final answers in the correct number of significant figures and with the correct units

1. A 7.546 g sample of an unknown compound was analyzed and found to contain only nitrogen and oxygen. If the mass of nitrogen present in the sample was 1.957 g, determine the empirical formula of the compound.
2. Serotonin is a compound that conducts nerve impulses in the brain. Serotonin has the following percent composition by mass: 68.2% carbon, 6.86% hydrogen, 15.9% nitrogen, and 9.08% oxygen.
 - a. What is its empirical formula?
 - b. What is its molecular formula if its molar mass is about 176 g/mol?
3. The koala dines exclusively on eucalyptus leaves. Its digestive system detoxifies the eucalyptus oil that can poison other animals. The chief constituent in eucalyptus oil is a substance called eucalyptol, which has the following mass percent composition: 77.87% carbon, 11.76% hydrogen, and the remainder is oxygen.
 - a. What is its empirical formula?
 - b. What is its molecular formula if its molar mass is about 154 g/mol?
4. Determine the molecular formulas for the following compounds given their empirical formulas and their molar masses: a) isoamyl formate: empirical formula= $\text{C}_3\text{H}_6\text{O}$, molar mass=116 g/mol; b) lactic acid: empirical formula= CH_2O , molar mass=90.1 g/mol; and c) opionic acid: empirical formula= $\text{C}_7\text{H}_2\text{O}$, molar mass=210.2 g/mol.

Empirical Formula Packet

MTG Learning Media

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Empirical Formula Packet:

CliffsNotes Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level *MCAT Study Review Notes - MEGA PACK 900 Pages* E Staff, MCAT Study Review Notes MEGA PACK 900 Pages Prepare for the MCAT with this review notes mega pack 900 pages of best review notes from successful MCAT test takers Know all the important facts that you need to succeed on the MCAT From quick facts and mnemonics and everything in between is included in this mega pack Review all the important areas of science Be prepared to ace the test and get admitted into a medical school Content created by highly successful former MCAT test takers with in depth knowledge of what it takes to succeed in this exam *Regents Chemistry--Physical Setting Power Pack Revised Edition* Albert S. Tarendash, 2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Chemistry Regents exam This edition includes Regents Exams and Answers Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day Let's Review Regents Chemistry Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key NASA Technical Note, 1973

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