

Empirical Formula, Molecular Formula and Hydrates Worksheet

1. Write the Empirical Formula for Each of the Following:

a. P_4O_6 _____

b. C_6H_9 _____

c. CH_2OHCH_2OH _____

d. $BrCl_2$ _____

e. $C_6H_8O_5$ _____

f. $C_{10}H_{22}$ _____

g. $Cu_2C_2O_4$ _____

h. Hg_2F_2 _____

2. Write the empirical formula for each of the following (show your work):

a. A compound composed of: 72% iron (Fe) and 27.6% oxygen (O) by mass. _____

b. A compound composed of: 9.93% carbon (C), 58.6% chlorine (Cl), and 31.4% fluorine (F).
_____ (This compound is commonly known as Freon)

c. A compound composed of: 0.556g carbon (C) and 0.0933g hydrogen (H). _____

3. Write the molecular formula for each of the following: (amu – atomic mass units, for our purposes it is equivalent to molecular mass of the compound).

a. A compound with a molecular mass of 70.0 amu and an empirical formula of CH_2 .

b. A compound with a molecular mass of 46.0 amu and an empirical formula of NO_2 .

4. Can the molecular formula of a compound ever be the same as the empirical formula for the compound? Explain your answer.

Empirical Molecular Formula Worksheet Answers

Jin-Ying Zhang



Empirical Molecular Formula Worksheet Answers:

General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Objective Workbook for Simplified ICSE Chemistry , Basics of Analytical Chemistry and Chemical Equilibria Brian M. Tissue, 2023-03-02

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Using ICT to Enhance Teaching and Learning in Chemistry Steve Lewis, 2004 Information and Communications Technology has revolutionised the ways we process access and use information and as computers and other devices become ever more powerful and information becomes more readily available the next generation will need to be able to interact with digital media effectively to exploit these amazing new technologies to their full potential for the benefit of society It is therefore imperative that teachers become familiar with ICT and its true potential and can present information with a perspective similar to that which the present generation of young people is using to develop their interests in their everyday life This resource from the RSC gives teachers of chemistry the practical help needed to integrate ICT into their teaching and stimulate the enthusiasm of a new generation of scientists in the exciting new areas of chemistry that are opening up such as Nanoscience and Nanotechnology Furthermore it will be highly effective in developing the new ethos of sustainability that will be a major driving force behind the next advances in chemistry that are vital if we are to survive the manifold problems confronting society in the next few decades This resource is ideal for all secondary teachers of chemistry trainee teachers and university lecturers

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Physicians' Desk Reference, 2005 Solving

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Formulas, Moles and Reactions Alessio Mangoni,2025-05-28 This comprehensive and rigorous educational text is dedicated to the study and application of the fundamentals of quantitative chemistry Through sixty clearly and precisely solved exercises it addresses essential topics such as molar mass and molecular weight the concept of the mole Avogadro s number the calculation of atoms and molecules in known quantities percent composition of compounds determination of empirical and molecular formulas and the balancing of chemical equations It serves as an effective tool for chemistry students and for anyone wishing to deepen their understanding of the discipline through a practical and methodical approach providing solid skills that are essential for continuing in scientific studies

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