lame:	Date:	Class:



## History of the Atom Timeline





#### John Dalton (1766 - 1844):

John Dalton was an English chemist. His ideas form the atomic theory of matter. Here are his ideas.

- All elements are composed (made up) of atoms. It is impossible to divide or destroy an atom.
- All atoms of the same elements are alike. (One atom of oxygen is like another atom of oxygen.)
- Atomic Theory

Dalton's

Modern.

- Atoms of different elements are different. (An atom of oxygen is different from an atom of hydrogen.)
- Atoms of different elements combine to form a compound. These atoms have to be in definite whole number ratios.
   For example, water is a compound made up of 2 atoms of hydrogen and 1 atom of oxygen (a ratio of 2:1). Three atoms of hydrogen and 2 atoms of oxygen cannot combine to make water.
- 1. What is the name of John Dalton's theory? THE ATOMIC THEORY OF MATTER
- 2. What are elements made of? ELEMENTS ARE MADE OF ATOMS
- 3. An atom of hydrogen and an atom of carbon are DIFFERENT
- 4. What are compounds made of? COMPOUNDS ARE MADE OF ATOMS OF TWO OR MORE DIFFERENT ELEMENTS
- 5. The ratio of atoms in HCl is: a) 1:3 b) 2:1 c) 1:1
- 6. The ratio of atoms in H<sub>2</sub>O is: a) 1:3 b) 2:1 c) 1:1 2 atoms of hydrogen and 1 atom of oxygen

# **History Of The Atom Worksheet Answers**

**Philip Roden** 

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industries for example as well as agriculture all utilize the science of chemistry This book helps everyone to understand nature However one does not need to be a chemist or scientist to understand the simplicity within the complexity around us The aim was to write a modern up to date book where students and teachers can get concise information about the structure of substances Sometimes reactions are given in the detailed form but in general excessive detail has been omitted The book is designed to introduce fundamental knowledge in three areas the history of the atom the periodic table and radioactivity We will study the historical development of atomic structure theories the tendencies of elements in periods and groups and the types of emissions and uses of radioactivity

From Atomos to Atom Andreas Gerardus Maria van Melsen,1960

History of the Atomic Theory John Williams Langley, 1878 A Short History of Atomism Joshua Craven Gregory, 1931 History of Atomic Theory George Parris, 2019-04-18 Most people have been exposed to the highlights of atomic theory But even chemists and physicists who have taken courses from modern text books only see the tip of the iceberg Indeed I believe that the textbooks themselves often distort the story Here I have tried to bring a higher level of detail and avoid some of the popular misconceptions about atomic theory and the people who developed it I think that these details assist in the understanding of the theory and the process of scientific discovery The book begins in 1800 and essentially ends in 1950 Beyond 1950 the developments in particle physics are actually a different activity than study of the atom per se and I have some doubts that the models of subatomic particles that have been produced over the last 50 years will stand the test of time In particular I find the history of the neutrino to be something of a distraction and mystery I have written about this issue before and provide an appendix that raises my concerns A New View of the Origin of Dalton's Atomic Theory: A Contribution to Chemical History (1896) Henry Enfield Roscoe, Arthur Harden, 2008-06 This scarce antiquarian book is a facsimile reprint of the original Due to its age it may contain imperfections such as marks notations marginalia and flawed pages Because we believe this work is culturally important we have made it available as part of our commitment for protecting preserving and promoting the world's literature in affordable high quality modern editions that are true to the A New View of the Origin of Dalton's Atomic Theory Henry Enfield Roscoe, Arthur Harden, 1896 original work

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