

## UNIT (6) ACIDS AND BASES

### 6.1 | Arrhenius Definition of Acids and Bases

Definitions for acids and bases were proposed by the Swedish chemist *Savante Arrhenius* in 1884.

**Acids** were defined as compounds that contain hydrogen and can dissolve in water to release hydrogen ions into solution. For example, HCl dissolves in water as follows:



**Bases** are defined as substances that dissolve in water to release hydroxide ions ( $\text{OH}^-$ ) into solution. For example, a typical base according to the Arrhenius definition is sodium hydroxide, NaOH:



The process that converts a molecule such as HCl into ions is called ionization.

**Ionization** is the production of ions from molecular compounds. No ions were initially present.

The process of converting a solid ionic compound, such as NaOH, into ions in solution is called dissociation.

**Dissociation** is the production of ions from ionic compounds. Ions were initially present, water just breaks apart ions.

**Some common acidic and basic substances:**

Acidic	Basic
citrus fruits	soap
vinegar	detergents
aspirin	milk of magnesia

**Some characteristic properties associated with acids and bases:**

Acids	Bases
sour taste	bitter taste
change litmus paper to red	change litmus paper to blue
react with hydroxide bases to produce salt	slippery, soapy feeling

# Harcourtschoolsupply Answers Unit 1

**Raffaela Di Napoli**



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