Test Unit 3 SCH3U #10

Name:

A Assessment of Learning (Product) 1.5

Multiple Choice: Circle the most correct answer. (10K)

What is the molar mass of AuCL?

A1 96 et

B) 130 a

(T) 232 S (c)

Do 303.6 @

Avogadro's number of representative particles is equal to one _____

A) kilogram B) gram

C) Kelvin

D0 mode.

3. How many moles of tungsten atoms are in 4.8 × 10²⁵ atoms of tunesten?

 $A) B 0 \times 10^{2}$ moles $B) B 0 \times 10^{4}$ moles $C) 1.3 \times 10^{-4}$ moles $D) 1.3 \times 10^{-9}$ moles

Determine the mass in grams of 0.0489 mol cobalt.

A) 0.00812 g B) 0.294 g C) 2.88 g

D) 0.000830 g

What is the mass in grams of 1.02 x 10²⁴ atoms manganese (Mn)?

A) 0.112 x 10° g

B) 0.169 x 10° g

C) 9.30 x 10⁻¹ g

D) 9.30 x 10° g.

6. Which of the following is/are true?

i. Samples of water coming from different sources will still have the same percentage composition

ii. The percentage composition of a compound can ONLY be calculated with knowledge of the compounds formula

iii. Percent composition is the relative volume of each element in the compound

iv Two compounds can have identical percentage compositions

A) L III

BY 1, 16, 166

C) II, III.

DO B. By

E) by only.

7. Which of the following is a candidate for an empirical formula?

A) CaHaOa B) NoOa

C) CH-OH D) C.H.

E) Na.S-O.

8. The mass percent of nitrogen, in nitric acid, HNO is

A) 1.6%

B) 11.1%

C) 15.6%

D) 22.2%

 Hydrogen peroxide is known to contain 5.9% hydrogen and 94.1% oxygen, by mass. If its molar mass. is 34.0 g/mol, its molecular formula is

AVEIO.

B) H-O.

C) HO.

DV HION

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