

Assessment

Quiz

Section: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best answers the question.

1. The formation of a gas is evidence of
 - a. a chemical change.
 - b. a physical change.
 - c. no change in energy.
 - d. Both (a) and (c).

2. Which of the following may be evidence of a chemical change?
 - a. release of energy as heat
 - b. production of sound
 - c. reduction in temperature
 - d. All of the above

3. In the word equation,

$$\text{iron} + \text{copper(II) sulfate} \rightarrow \text{iron(II) sulfate} + \text{copper}$$
 a product is
 - a. iron(II) sulfate.
 - b. copper.
 - c. iron.
 - d. Both (a) and (b)

4. In the reaction described by the following equation

$$\text{water} + \text{energy} \rightarrow \text{hydrogen gas} + \text{oxygen gas}$$
 energy
 - a. can be considered a reactant.
 - b. is absorbed.
 - c. is released.
 - d. Both (a) and (b)

5. In the unbalanced formula equation

$$\text{carbon} + \text{hydrogen} \rightarrow \text{methane}$$
 the formula CH_4 represents
 - a. one atom of methane.
 - b. five atoms of methane.
 - c. one molecule of methane.
 - d. one gram of methane.

Holt Chemistry Section One Assessment Quiz Answers

Kathleen Armour



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