

## QUESTION 2

|  | Test                                                    | Observation                                                                                                         | Inference                         |
|----------------------------------------------------------------------------------|---------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------|-----------------------------------|
| (a)(i)                                                                           | C + Conc. $H_2SO_4$ + strong heat                       | A copious brown irritating fume which turned moist blue litmus paper to red                                         | Gas is $NO_2$<br>$NO_3^-$ present |
| (b)(i)                                                                           | C + water + NaOH in drops, in excess                    | white gelatinous precipitate<br>precipitate dissolved                                                               | $Al^{3+}$ or $Zn^{2+}$            |
| (ii)                                                                             | C + water + $NH_3$ in drops, in excess                  | white gelatinous precipitate<br>precipitate dissolved                                                               | $Al^{3+}$ or $Zn^{2+}$            |
| (iii)                                                                            | C + water + freshly prepared $FeSO_4$ + Conc. $H_2SO_4$ | A brown ring is formed at the junction of the acid and aqueous solution.                                            | $NO_3^-$                          |
| (c)(i)                                                                           | D + water                                               | D dissolved                                                                                                         | soluble                           |
| (ii)                                                                             | Solution D + dil HCl                                    | Colourless gas, smells like rotten egg, turns moist blue litmus paper to red and lead (II) ethanoate paper to black | $H_2S$ present                    |
| (iii)                                                                            | Solution D + NaOH in drops, in excess                   | white precipitate<br>precipitate insoluble                                                                          | $Ca^{2+}$ or $Pb^{2+}$            |
| (iv)                                                                             | solution D + $(NH_4)_2CO_3$                             | white precipitate                                                                                                   | $Ca^{2+}$                         |

Trickle slowly down

# Gce 2013 Practical Chemistry Answers

**Elizabeth Preston**



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## **Table of Contents Gce 2013 Practical Chemistry Answers**

1. Understanding the eBook Gce 2013 Practical Chemistry Answers
  - The Rise of Digital Reading Gce 2013 Practical Chemistry Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Gce 2013 Practical Chemistry Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gce 2013 Practical Chemistry Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gce 2013 Practical Chemistry Answers
  - Personalized Recommendations
  - Gce 2013 Practical Chemistry Answers User Reviews and Ratings

- Gce 2013 Practical Chemistry Answers and Bestseller Lists
- 5. Accessing Gce 2013 Practical Chemistry Answers Free and Paid eBooks
  - Gce 2013 Practical Chemistry Answers Public Domain eBooks
  - Gce 2013 Practical Chemistry Answers eBook Subscription Services
  - Gce 2013 Practical Chemistry Answers Budget-Friendly Options
- 6. Navigating Gce 2013 Practical Chemistry Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Gce 2013 Practical Chemistry Answers Compatibility with Devices
  - Gce 2013 Practical Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Gce 2013 Practical Chemistry Answers
  - Highlighting and Note-Taking Gce 2013 Practical Chemistry Answers
  - Interactive Elements Gce 2013 Practical Chemistry Answers
- 8. Staying Engaged with Gce 2013 Practical Chemistry Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gce 2013 Practical Chemistry Answers
- 9. Balancing eBooks and Physical Books Gce 2013 Practical Chemistry Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gce 2013 Practical Chemistry Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Gce 2013 Practical Chemistry Answers
  - Setting Reading Goals Gce 2013 Practical Chemistry Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gce 2013 Practical Chemistry Answers
  - Fact-Checking eBook Content of Gce 2013 Practical Chemistry Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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