

How Substances Dissolve

KEY IDEAS

As you read this section, keep these questions in mind:

- Why is water called the universal solvent?
- How do substances dissolve?

Why Do Substances Dissolve in Certain Solvents?

Water is often called a universal solvent because many substances dissolve in it. However, no one substance can dissolve every solute. A general rule in chemistry is that “like dissolves like.” This rule means that a solvent will dissolve substances that have similar molecular structures. 

A solvent must be able to attract solute particles and pull them away from one another. Thus, solvent particles must attract solute particles more strongly than the solute particles attract one another.

POLAR MOLECULES

Recall that atoms bonded together do not all share electrons equally. For example, a water molecule is made up of two hydrogen atoms bonded to one oxygen atom. The oxygen atom attracts electrons more strongly than the hydrogen atoms do. Thus, the electrons spend more time near the oxygen atom. This gives each hydrogen atom a partial positive charge (δ^+) and the oxygen atom a partial negative charge (δ^-). 

Molecules that have partially charged positive and negative areas are **polar**. Because water molecules are polar, water can dissolve many other polar substances. Water can also dissolve ionic compounds such as table salt.

Water is a polar molecule because the oxygen atom strongly attracts electrons. Because the electrons spend more time near the oxygen atom, the oxygen atom has a partial negative charge.



READING TOOLBOX

Summarize Read each page of the section silently to yourself. With a partner, take turns summarizing each page.

READING CHECK

1. Identify Why is water called a universal solvent?

because
many substances dissolve
in it

READING CHECK

2. Explain Why does a water molecule have partial charges?

The oxygen atom attracts
electrons more strongly
than the hydrogen atoms
do. Thus, the electrons
spend more time near the
oxygen atom.

How Substances Dissolve Concept Review Answers

**David Sang, Peter Ellis, Derek
McMonagle**



How Substances Dissolve Concept Review Answers:

Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change Chemistry in the Community. American Chemical Society, 2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts *Scientifica* David Sang, Peter Ellis, Derek McMonagle, 2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities **Science I Essential Interactions** , 2000-10 **MCAT General Chemistry Review** Princeton Review (Firm), 2010 The MCAT is a test of more than just the facts about basic physical and biological sciences it s an in depth rigorous examination of your knowledge of scientific concepts and principles as well as your critical thinking and writing skills With the Princeton Review s subject specific MCAT series you can focus your review on the MCAT topics that are most challenging to you Each book in the series contains the most in depth coverage of subjects tested on the MCAT Each chapter in MCAT General Chemistry includes Full color illustrations charts and diagrams Examples of general chemistry questions and their solutions worked out step by step Chapter Review Quizzes and answers A real MCAT style practice passage with questions and answers Bulleted chapter summaries for quick review MCAT General Chemistry Review also includes A complete glossary of general chemistry terms A general chemistry formula sheet *General Science* Richard H. Moyer, 1986 **Pharmaceutical Record and Weekly Market Review** P. W. Bedford, 1892 *Chemistry: Core Concepts, 3rd Edition* Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie Williamson, Christopher Thompson, 2024-01-09 The third edition of Chemistry Core Concepts Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package **Nals Gr1:english Teachers Gde** , **Concepts and Challenges in Science New York Proficiency Review Book** Globe Fearon, 2000-11 *International Review of Cytology* , 1976-10-20 International Review of Cytology **MCAT Organic Chemistry Review** Kaplan Test Prep, 2016-07-05 More people get into medical school with a Kaplan MCAT course than all major courses combined Now the same results are available with MCAT Organic Chemistry Review This book features thorough subject review more questions than any competitor and the highest yield questions available The commentary and instruction come directly from Kaplan MCAT experts and include targeted focus on the most tested concepts MCAT Organic Chemistry Review offers UNPARALLELED MCAT KNOWLEDGE The Kaplan MCAT team has spent years studying every MCAT related document available In conjunction with our expert psychometricians the Kaplan team is able to ensure the accuracy and realism of our practice materials THOROUGH SUBJECT REVIEW Written by top rated award

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Positivist Review, 1903

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Basic Concepts of Chemistry, 9e

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