

What Does It Mean to Say "The Temperature of the System is Constant"?

Suppose you have a system of particles (e.g., a gas) in a container.

Consider now a process where the system is heated. The temperature of the system is constant. This means that the average kinetic energy of the particles is constant. This is only true if the system is in thermal contact with a reservoir.

Now, the "reservoir" (which is a large system) is assumed to be at a constant temperature. This means that the average kinetic energy of the particles in the reservoir is constant. This is only true if the reservoir is in thermal contact with a larger reservoir.

Therefore, the temperature of the system is constant only if the system is in thermal contact with a reservoir that is in thermal contact with a larger reservoir. This is only true if the system is in thermal contact with a reservoir that is in thermal contact with a larger reservoir.

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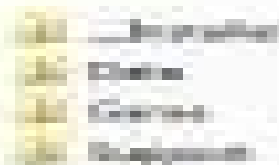
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