## CBSE Science - Grade X Solution for 2022-23 Examination

Tooth enamel starts corroding in case the pH becomes acidic. The germs in the mouth decompose the food stuck in between the teeth and produce acids. These acids react with the enamel causing tooth decay. Tooth decay can be prevented by using toothpaste for brushing teeth and keeping oral hygiene.

#### 33.

(a) Identify the reducing agent in the following reactions

- (i) 4NH<sub>3</sub> + 5O<sub>2</sub> → 4NO + 6H<sub>2</sub>O
- (ii) H<sub>2</sub>O + F<sub>2</sub> → HF + HOF
- (iii)  $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$
- (iv)  $2H_2 + O_2 \rightarrow 2H_2O$ 
  - (b) Define a redox reaction in terms of gain or loss of oxygen.

#### Answer: (a)

The substance that oxidized is the reducing agent and the substance which gets reduced is the oxidizing agent.

(i) In the reaction 4NH<sub>3</sub> + 5O<sub>2</sub> → 4NO + 6H<sub>2</sub>O

There is the oxidation of ammonia (NH<sub>3</sub>) and reduction of oxygen (O<sub>2</sub>) taking place so here NH<sub>3</sub> is the reducing agent and O<sub>2</sub> is oxidising agent.

(ii) In the reaction H<sub>2</sub>O + F<sub>2</sub> → HF + HOF

There is the oxidation of  $H_2O$  and reduction of  $F_2$  taking place so here  $H_2O$  is the reducing agent and  $F_2$  is oxidising agent.

(iii) In the reaction Fe<sub>2</sub>O<sub>3</sub> + 3CO → 2Fe + 3CO<sub>2</sub>

There is the oxidation of CO and reduction of Fe<sub>2</sub>O<sub>3</sub> taking place so here CO is the reducing agent and Fe<sub>2</sub>O<sub>3</sub> is oxidising agent.

(iv) In the reaction 2H₂ + O₂ → 2H₂O

There is the oxidation of  $H_2$  and reduction of  $O_2$  taking place so here  $H_2$  is the reducing agent and  $O_2$  is oxidising agent.

(b) Oxidation: A chemical reaction in which a substance gains oxygen is called oxidation.

Example: 2Cu + O<sub>2</sub> = 2CuO (Cu is oxidized to CuO)

# **March 2nengineering Science Question Paper**

**Daniela Niemeyer** 

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