

13. Run alongside the test variables at all times
 14. **Not using specific reagents are valid for this purpose**
15. **False** positive pregnancy results from the test may mean:
 a) The patient is actually pregnant
 b) **The patient is not pregnant but has been given a wrong result**
 c) The reagents used are very specific and accurate
 d) The patient had taken a lot of protein
16. Urine with high specific gravity may indicate any of the following **EXCEPT**:
 a) Dehydration
 b) Glycosuria
 c) **None of the patient**
 d) Proteinuria
17. Ketones in urine are detected by:
 a) **Rose**
 b) Barium
 c) Clinette
 d) Clinchem

SECTION B: ESSAY QUESTIONS (20 MARKS)

1. a) Given the patients urine report as TDS mg/dL, convert this into blood urea nitrogen (BUN).
 (B_u = 14, G_u = 16, G_u = 12, B_u = 7) **(2 marks)**
- b) If 0.5 g of CaCl₂ are made to 1 liter, what is the concentration of the solution in terms of molarity and percentage (w/v)?
 (Ca = 40, Cl = 35.5) **(2 marks)**
2. Define and describe the role of the following in quantitative analytical methods:
 (a) Blank solution
 (b) Standard solution
 (c) Control solution
 (d) Reference ranges **(10 marks)**
3. a) List in **CONNECT** sequence, the five main components of a single colorimeter **(5 marks)**
- b) List any three advantages of quality control in a clinical chemistry laboratory **(3 marks)**

Immunology

- (1) describe the following processes and give their uses in testing:
 (a) zymic acid (b) PCA, (c) fluorescent oligonucleotide assay
 (2) give the advantages and reasons of freezing tissues
 (3) describe the process of decalcification

Clinical chemistry

- (1) differentiate between the following:-

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Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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