

TSBIE - MODEL PAPER CHEMISTRY - I

Time: 3 Hours

Max.Marks: 60

Each Question Carries TWO marks.

2 x 10 = 20

SECTION - A

1. What is Lanthanide Contraction? Give one of its consequences.
2. Which of the two ions Ca^{2+} or Zn^{2+} is more stable and why?
3. What is Boltzman's constant? Give its value.
4. Calculate the weight of 0.1 mole of sodium carbonate.
5. What is homogenous equilibrium? Write two homogenous reactions.
6. Lithium salts are mostly hydrated. Why?
7. What happens when magnesium metal is burnt in air?
8. Diamond has high melting point – explain.
9. What is the effect of water on tin (Sn)?
10. Write the structure of a) Neopentane b) Trichloro ethanoic acid.

SECTION - B

Each Question Carries FOUR marks.

6 x 4 = 24

11. Explain the differences between emission and absorption spectrum.
12. Balance the following reaction by ion electron method.
$$\text{Cr}_2\text{O}_7^{2-} + \text{SO}_3^{2-} \xrightarrow{\text{H}^+} \text{Cr}^{3+} + \text{SO}_4^{2-}$$
13. Write the postulates of Kinetic Molecular theory of gases.
14. Derive Ideal gas equation.
15. Chemical analysis of a carbon compound gave the following percentage composition by weight of the elements present, carbon = 10.06%, hydrogen = 0.84%, chlorine = 89.10%, Calculate the empirical formula of the compound.
16. State and explain Hess's law of constant heat summation.

Intermediate 1st Year Chemistry Model Papers 23

JA Banks



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Table of Contents Intermediate 1st Year Chemistry Model Papers 23

1. Understanding the eBook Intermediate 1st Year Chemistry Model Papers 23
 - The Rise of Digital Reading Intermediate 1st Year Chemistry Model Papers 23
 - Advantages of eBooks Over Traditional Books
2. Identifying Intermediate 1st Year Chemistry Model Papers 23
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Intermediate 1st Year Chemistry Model Papers 23
 - User-Friendly Interface
4. Exploring eBook Recommendations from Intermediate 1st Year Chemistry Model Papers 23
 - Personalized Recommendations
 - Intermediate 1st Year Chemistry Model Papers 23 User Reviews and Ratings
 - Intermediate 1st Year Chemistry Model Papers 23 and Bestseller Lists
5. Accessing Intermediate 1st Year Chemistry Model Papers 23 Free and Paid eBooks

- Intermediate 1st Year Chemistry Model Papers 23 Public Domain eBooks
- Intermediate 1st Year Chemistry Model Papers 23 eBook Subscription Services
- Intermediate 1st Year Chemistry Model Papers 23 Budget-Friendly Options
- 6. Navigating Intermediate 1st Year Chemistry Model Papers 23 eBook Formats
 - ePub, PDF, MOBI, and More
 - Intermediate 1st Year Chemistry Model Papers 23 Compatibility with Devices
 - Intermediate 1st Year Chemistry Model Papers 23 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Intermediate 1st Year Chemistry Model Papers 23
 - Highlighting and Note-Taking Intermediate 1st Year Chemistry Model Papers 23
 - Interactive Elements Intermediate 1st Year Chemistry Model Papers 23
- 8. Staying Engaged with Intermediate 1st Year Chemistry Model Papers 23
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Intermediate 1st Year Chemistry Model Papers 23
- 9. Balancing eBooks and Physical Books Intermediate 1st Year Chemistry Model Papers 23
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Intermediate 1st Year Chemistry Model Papers 23
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Intermediate 1st Year Chemistry Model Papers 23
 - Setting Reading Goals Intermediate 1st Year Chemistry Model Papers 23
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Intermediate 1st Year Chemistry Model Papers 23
 - Fact-Checking eBook Content of Intermediate 1st Year Chemistry Model Papers 23
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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