

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

সংকেত, $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে দ্রবিত X-পদার্থিতে 10 mg এর মধ্যে কয় পরিমাণ দ্রব
যে X-পদার্থিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl দ্রব দিতে কাজ করা হয়েছে।

যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও
Y-পদ্ধতি হলো ছোট আকারে আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির
মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা
হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
বিশেষতঃ বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05
থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2S এর পরিবর্তে
কোয়েলিনাইট ব্যবহার করা হয় ফলে পরিবেশ দূষণ ঘটে না।
 $(\text{CNH}_2)_2 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{CONH}_2 + \text{H}_2\text{S}$

অতএব, কোনো কোনো ক্ষেত্রে কম ঝুঁকির রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

উদাহরণ



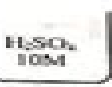
পাত্র-ক



পাত্র-খ



পাত্র-গ



পাত্র-ঘ

১. টাইটার দ্রবক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঙ্কার্ড প্রতীকটি অঙ্কন করো এবং ব্যাখ্যা করো।
৪. উদাহরণের অংশের দ্রবণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করো। ৪

৪ নং প্রশ্নের উত্তর

১. বিশ্লেষণের নিমিত্ত বীমের উপর টাইটার স্থাপন করলে বীমের প্রতি
দিকের ভর সমান হলে টাইটার দ্রবক বলে।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
বিশেষতঃ বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ
(0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।
অতএব, কোনো কোনো ক্ষেত্রে কম ঝুঁকির রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

৩. উদাহরণের দ্রবণটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M
দ্রব। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M
দ্রব যখন পাত্র H_2SO_4 দ্রবণ। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঙ্কার্ড
প্রতীক অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঙ্কার্ড
প্রতীক নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আবশ্য	ক্ষয়কারী রাসায়নিক উপাদান কলতে যে ক্ষয় রাসায়নিক দ্রব্যকে খুবই দ্রুত সংস্পর্শে তুলে ও অন্যান্য পদার্থের ক্ষয়কারী ক্ষতি সাধিত হয়। এতে তুলে পুড়ে যায় এবং অন্যান্য পদার্থ গলে যায়।
উদাহরণ	পাত্র কঠিন এসিডসলফুরিক (HCl, HNO_3 , H_2SO_4), পাত্র ক্ষয় (NaOH, KOH), H_2O_2 , অক্সিজেন লিকার ক্লোরিন দ্রবণ ইত্যাদি।
সতর্কতা সংকেত	- পাত্র এসিডকে সোতলে রাখতে হবে। - পাত্র সংরক্ষণে, এরপর স্পর্শের পর সতর্কতা সতর্কতা রাখতে হবে।

উদাহরণের দ্রবণটি হলো 10M H_2SO_4 দ্রবণ এবং এর থেকে 1M
ঘনমাত্রার দ্রবণ প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে
হবে।

1ম চলকের দ্রবণ হলো 100 ml অক্সিজেনমুক্ত জল ও এর 10 ml
দ্রবণ। সেমি-মাইক্রো পদ্ধতির দ্রবণের আয়তন 100 ml। তাই
আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রবণ তৈরি করতে হবে।
দ্রবণের গণনা দ্রবণ দিতে জানতে পারি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 দ্রবণের সঙ্গে পানি যোগ করতে হবে
(100-10) mL = 90 mL
আবার, দ্রবণটির সত্যতা 10 ml দ্রবণ ঘনমাত্রার দ্রবণ নিয়ে তার
সঙ্গে অক্সিজেনমুক্ত জলকে 90 ml পানি যোগ করলে আমাদের
কামিষ্ট ঘনমাত্রার দ্রবণ পাওয়া যাবে।

উদাহরণ



চিত্র-A



চিত্র-B



চিত্র-C

১. অক্সিজেনমুক্ত জলকে বসে।
২. সমান দ্রবক K, এর মান কখনো অসীম হয় না-ব্যাখ্যা
করো।
৩. A চিত্রের হ্যাঙ্কার্ড উত্তর করার কৌশল বর্ণনা করো।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক দ্রবণের
ব্যবহারকালে খান্সা ঝুঁকি হ্রাসে কবীর উপায় সম্পর্কে
বিশ্লেষণ করো। ৪

৪ নং প্রশ্নের উত্তর

১. অক্সিজেনমুক্ত জল এক ধরনের রাসায়নিক বা কৃত্রিম দ্রবণ সতর্কত
বা মৃত্যু বা অন্য উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে তুলে
খাবার সতর্কত করে।

২. সামান্য K, এর মান কখনো অসীম বা শূন্য হতে পারে না।
উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণ মাইক্রোপদ্ধতি বিশ্লেষণ করি-
 $A + B \rightarrow C + D$

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RM Cervero



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