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following mechanism
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ (fast)
 $\text{O} + \text{O}_2 \rightarrow \text{O}_3$ (slow)
 a) write the overall balanced chemical equation
 $2\text{O}_3 \rightarrow 3\text{O}_2$
 b) identify any intermediates within the mechanism
 c) what is the order with respect

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