

Griddable Science Items

Topography map

Density

Atoms

$$F = ma$$

			6	.	2	5
0	0	0	0		0	0
1	1	1	1		1	1
2	2	2	2		2	2
3	3	3	3		3	3
4	4	4	4		4	4
5	5	5	5		5	5
6	6	6	6		6	6
7	7	7	7		7	7
8	8	8	8		8	8
9	9	9	9		9	9

Designed for

8th grade STAAR Test



but would work for other standardized tests

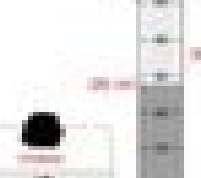
Cricketables

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1. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 16.7 g and a volume of 4.2 cm³. What is the density of the mineral sample in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{16.7}{4.2} = 3.98 \text{ g/cm}^3$$

2. A rock is 1.2 cm³. What is the density of the rock in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{1.2}{1} = 1.2 \text{ g/cm}^3$$

3. A sample of metal has a mass of 25 g. What is the density in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{25}{1} = 25 \text{ g/cm}^3$$

Name: _____ ANSWERS

1. What is the density of the liquid in the graduated cylinder?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{4.00}{10.00} = 0.40 \text{ g/mL}$$

2. The sample of metal has a mass of 25 g. What is the density in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{25}{5 \times 5 \times 1} = 1 \text{ g/cm}^3$$

Answer your questions about the problems for each to use in the next step.

1. 

2. 

3. 

4. 

5. 

6. 

81

ST

Gridable Science 8th Grade Staar Practice

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