

IT Ramalah SAT Model Question Paper

CHEMISTRY

Time : One Hour

MAX Marks : 50

Instructions:

- Answer must be written either in English or the medium of instruction of the candidate in high school.
- There will be no negative marking.
- Use of calculators or graph papers is not permitted.

Answer all the questions. Each question carries 2½ marks.

- Naphthalene is soluble in benzene but not in water. Give reason.
- Explain the following :
 - Aqueous solution of ammonium acetate is almost neutral.
 - Both coke and flux are used in smelting process of metallurgy.
- pH of 10^{-4} M HCl is not equal to 4.
- Differentiate the following :
 - 1-propanol and 2-propanol.
 - Acid strength and concentration.
- For the reaction

$$(i) \text{A} \rightleftharpoons \text{B} \quad K_c = 2$$

$$(ii) \text{B} \rightleftharpoons \text{C} \quad K_c = 4$$

$$(iii) \text{C} \rightleftharpoons \text{D} \quad K_c = 6$$
 For the reaction $\text{A} \rightleftharpoons \text{D}$ what is the K_c ?
- A fuel gas contains carbon and hydrogen only. Burning a small sample of it in oxygen gives 2.36gm carbon dioxide,

8.84gm of water and no other products. A volume of 10 l (measured at STP) of this gas is found to weigh 11.4gm. Calculate (i) empirical formula (ii) molecular mass of the gas (iii) molecular formula.

- Why hydrogen peroxide acts as oxidising as well as reducing agent ?
- Differentiate the following :
 - Hard water and heavy water.
 - Lattice energy and hydration energy.
- What is the action of heat on following salts? Explain with balanced equations

$$(i) \text{FeSO}_4 \xrightarrow{\Delta} \quad \quad \quad (ii) \text{AgNO}_3 \xrightarrow{\Delta}$$
- What is electrolyte ? What is the basic requirement for a substance to act as electrolyte ?
- Diamond and graphite are both made of carbon atom. Yet diamond does not conduct electricity while graphite does. Explain?
- When 280 g of washing soda crystals are left in dry air for some time, a loss of weight of 162 g occurs. How can you account for this ?
- Total hardness of water cannot be removed by simple heating. Why ?
- All exothermic reactions are not spontaneous reactions. Why ?
- Bottle containing NaOH solution should not be closed by glass lid. Explain?
- How is litmus indicator prepared?
- Cl_2 is nonpolar while HCl is polar. Explain?

- What are the internal factors that decide the physical state of the substance ?
- What is the total no. of protons, electrons and neutrons in 12gms of C_{12}^{12} isotopes?
- HNO_3 stains the skin yellow. Why?

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