

**ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পটীকার প্রস্তুতকরণ**

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$W = 20 \text{ g}$$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন,  $V = 1 \text{ L}$

চলকের ঘনমাত্রা  $S = 1$

উদাহরণে উল্লিখিত  $N$ -পদ্ধতিতে 10 mg এর মধ্যে কয় পরিমাণ ঘন ও  $V$ -পদ্ধতিতে 0.2 g 2%  $\left(\frac{W}{V}\right)$  HCl নমুনা নিয়ে কাজ করা হয়েছে।

যদিও  $N$ -পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও  $V$ -পদ্ধতিটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশন ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে  $H_2SO_4$  এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে মুদ্রণ ঘটে না।

$(CH_3NH_2 + H_2O \rightarrow CH_3NH_3^+ + OH^-)$  ফলে কোনো কোনো ক্ষেত্রে কম কষ্টিকর রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

**উদাহরণ**



পাত্র-ক



পাত্র-খ



পাত্র-গ



পাত্র-ঘ

১. টাইট্রার যুক্ত কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অংশের চলক থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার চলক তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ ও না প্রসঙ্গের উত্তর

১. বিশ্লেষণের নিমিত্ত বীমের উপর টাইট্রার স্থাপন করলে বীমের প্রতি পাঠকের জন্য যে ভর পাওয়া যায়, তাকে টাইট্রার যুক্ত বলে।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশন ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

৩. অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টিকর রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

প্রতীক	অর্থ (Corrosive)
আয়তন	অসংখ্য রাসায়নিক উপাদান চলকে যে মাত্রায় রাসায়নিক দ্রব্যকে যুক্ত করা হয় তাৎপর্যপূর্ণ ভুল ও অন্যান্য পদ্ধতির মাধ্যমে সঠিক পরিমিত হয়। এতে ভুল পুঙ্কে যায় এবং অন্যান্য পদার্থ পুঙ্কে যায়।
উপাদান	পাত্র বিনামূলি (HCl, HNO <sub>3</sub> , H <sub>2</sub> SO <sub>4</sub> ) পাত্র ফল (NaOH, KOH), H <sub>2</sub> O <sub>2</sub> , অ্যাসেনিফিক লিকার প্রভৃতি চলক ইত্যাদি।
সংরক্ষণ পাত্র/ভা	<ul style="list-style-type: none"> <li>পাত্র এম্বিকলে সোতলে রাখতে হবে।</li> <li>পাত্র ফলকে, ফলকে পাত্রের পৃষ্ঠ থেকে সোতলে রাখতে হবে।</li> </ul>

উদাহরণের চলকটি হলো 10M  $H_2SO_4$  চলক এবং এর থেকে 1M ঘনমাত্রার চলক প্রস্তুত করতে হবে 1ম ও ১ম চলকের ব্যবহার করতে হবে।

1ম চলকটির জন্য 100 ml অ্যাসেনিফিক চলক ও এর টি মাত্র পিঙ্কে। সেমি-মাইক্রো পদ্ধতির চলকের আয়তন 100 ml। তাই আমাদের 100 ml 1M ঘনমাত্রার  $H_2SO_4$  চলক তৈরি করতে হবে। চলকের সন্নিবেশন পুঙ্কে করে আনত।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left( \frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml  $H_2SO_4$  চলকের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL।

আবার, পিঙ্কের মাধ্যমে 10 ml চলকের ঘনমাত্রার চলক নিয়ে তার সাথে অ্যাসেনিফিক চলকে 90 ml পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার চলক পাওয়া যাবে।

**উদাহরণ**



চিত্র-A



চিত্র-B



চিত্র-C

১. অ্যাসেনিফিক কী বলে?
২. সাদা চলক K<sub>2</sub> এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিত্রের হ্যাঞ্জারে উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিত্রের মাধ্যমে সম্পর্কিত রাসায়নিক চলকের ব্যবহারকালে খাওয়া কষ্টিকর উপায় সম্পর্কে বিশ্লেষণ করে। ৪ ও না প্রসঙ্গের উত্তর

১. অ্যাসেনিফিক কী বলে? অসংখ্য রাসায়নিক বা কৃত্রিম দ্রব্য সোতল বা মুদ্রণ বা উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে রক্ষা করে রাখার সতর্কতা করে।

২. সামান্য K<sub>2</sub> এর ঘন কখনো অসীম বা শূন্য হতে পারে না। উত্তরটি ব্যাখ্যা করার জন্য নিম্নের উত্তরসূচী সঠিকরণটি বিশ্লেষণ করি-  
 $A + B \rightarrow C + D$

# Hsc Chemistry 1 First Chapter Notes

**Sebastian Brünink**



## Hsc Chemistry 1 First Chapter Notes:

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