

Simplifying Rational Exponents

Date _____ Period _____

Simplify.

1) $(n^4)^{\frac{5}{2}}$

2) $(27p^6)^{\frac{5}{3}}$

3) $(25b^6)^{-1.5}$

4) $(64m^4)^{\frac{3}{2}}$

5) $(a^8)^{\frac{1}{2}}$

6) $(9r^4)^{0.5}$

7) $(81x^{12})^{1.25}$

8) $(216r^9)^{\frac{1}{3}}$

Simplify. Your answer should contain only positive exponents with no fractional exponents in the denominator.

9) $2m^2 \cdot 4m^{\frac{3}{2}} \cdot 4m^{-2}$

10) $3b^{\frac{1}{2}} \cdot b^{\frac{4}{3}}$

11) $\left(p^{\frac{1}{2}}\right)^{-2}$

12) $\left(a^{\frac{5}{12}}\right)^{\frac{3}{5}}$

Kuta Simplifying Exponents Expressions

Mike Jess

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Kuta Simplifying Exponents Expressions:

Izvestiya, Russian Academy of Sciences ,1994 **Physics, Uspekhi** ,2004 **Journal of the Optical Society of America** ,1995 Comprehensive Treatise of Electrochemistry Ralph E. White,1984-09-30 It is now time for a comprehensive treatise to look at the whole field of electrochemistry The present treatise was conceived in 1974 and the earliest invitations to authors for contributions were made in 1975 The completion of the early volumes has been delayed by various factors There has been no attempt to make each article emphasize the most recent situation at the expense of an overall statement of the modern view This treatise is not a collection of articles from Recent Advances in Electrochemistry or Modern Aspects of Electrochemistry It is an attempt at making a mature statement about the present position in the vast area of what is best looked at as a new interdisciplinary field Texas A M University J O M Bockris University of Ottawa B E Conway Case Western Reserve University Ernest Yeager Texas A M University Ralph E White Preface to Volume 8 Experimental methods in electrochemistry are becoming more diverse This volume describes many of the new techniques that are being used as well as some of the well established techniques It begins with two chapters 1 and 2 on electronic instrumentation and methods for utilization of microcomputers for experimental data acquisition and reduction Next two chapters 3 and 4 on classical methods of electrochemical analysis are presented ion selective electrodes and polarography

Comprehensive Treatise of Electrochemistry: Experimental methods in electrochemistry John O'M. Bockris,1980
British Chemical Engineering & Process Technology ,1969 **British Chemical Engineering** ,1969

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