

Test Unit 3 SCH3U #10

Name: _____

15 K 20 T 5 C 7 A Assessment of Learning (Product)

Multiple Choice: Circle the most correct answer. (10K)

1. What is the molar mass of AuCl_3 ?

- A) 96 g B) 130 g C) 232.5 g D) 303.6 g

2. Avogadro's number of representative particles is equal to one _____.

- A) kilogram B) gram C) Kelvin D) mole

3. How many moles of tungsten atoms are in 4.8×10^{22} atoms of tungsten?

- A) 8.0×10^2 moles B) 8.0×10^1 moles C) 1.3×10^{-1} moles D) 1.3×10^{-2} moles

4. Determine the mass in grams of 0.0489 mol cobalt.

- A) 0.00812 g B) 0.294 g C) 2.88 g D) 0.000830 g

5. What is the mass in grams of 1.02×10^{24} atoms manganese (Mn)?

- A) 0.112×10^1 g B) 0.169×10^6 g C) 9.30×10^{-1} g D) 9.30×10^6 g

6. Which of the following is/are true?

- i. Samples of water coming from different sources will still have the same percentage composition
- ii. The percentage composition of a compound can ONLY be calculated with knowledge of the compound's formula
- iii. Percent composition is the relative volume of each element in the compound
- iv. Two compounds can have identical percentage compositions

- A) i, ii B) i, ii, iii C) ii, iii D) i, iv E) iv only

7. Which of the following is a candidate for an empirical formula?

- A) $\text{C}_{24}\text{H}_{40}\text{O}_{22}$ B) N_2O_4 C) CH_3OH D) C_6H_6 E) $\text{Na}_2\text{S}_2\text{O}_6$

8. The mass percent of nitrogen, in nitric acid, HNO_3 , is

- A) 1.6% B) 11.1% C) 15.6% D) 22.2%

9. Hydrogen peroxide is known to contain 5.9% hydrogen and 94.1% oxygen, by mass. If its molar mass is 34.0 g/mol, its molecular formula is

- A) HO B) H_2O_2 C) HO_2 D) HO_3

Grade 11 Stoichiometry Practice Test

Padhraic Smyth

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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A Stoichiometry Unit David Callaghan, 2004 **Understanding Physics Using Mathematical Reasoning** Andrzej Sokolowski, 2021-08-20 This book speaks about physics discoveries that intertwine mathematical reasoning modeling and scientific inquiry It offers ways of bringing together the structural domain of mathematics and the content of physics in one coherent inquiry Teaching and learning physics is challenging because students lack the skills to merge these learning paradigms The purpose of this book is not only to improve access to the understanding of natural phenomena but also to inspire new ways of delivering and understanding the complex concepts of physics To sustain physics education in college classrooms authentic training that would help develop high school students skills of transcending function modeling techniques to reason scientifically is needed and this book aspires to offer such training The book draws on current research in developing students mathematical reasoning It identifies areas for advancements and proposes a conceptual framework that is tested in several case studies designed using that framework Modeling Newton s laws using limited case analysis Modeling projectile motion using parametric equations and Enabling covariational reasoning in Einstein formula for the photoelectric effect represent some of these case studies A wealth of conclusions that accompany these case studies drawn from the realities of classroom teaching is to help physics teachers and researchers adopt these ideas in practice

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