

TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm³

Use only the student's concordant results

mean titre = $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$ ✓

$$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol} \quad \checkmark$$

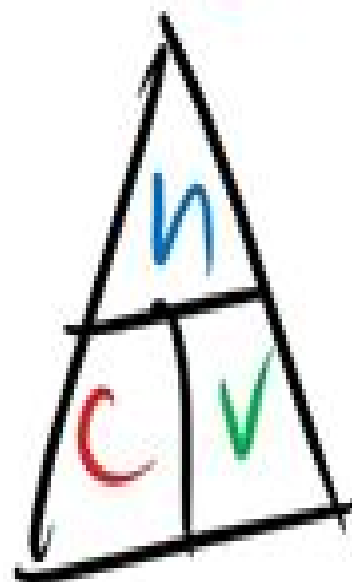
$$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol} \quad \checkmark$$

$$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3 \quad \checkmark$$

$$n = m/M_r$$

$$n = c \times v$$

$$c = n/v$$



GCSE CHEMISTRY

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Adrian Schmit, Jeremy Pollard

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the authors' names.

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Table of Contents Gcse Titration Practise Questions

1. Understanding the eBook Gcse Titration Practise Questions
 - The Rise of Digital Reading Gcse Titration Practise Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcse Titration Practise Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Titration Practise Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcse Titration Practise Questions
 - Personalized Recommendations
 - Gcse Titration Practise Questions User Reviews and Ratings
 - Gcse Titration Practise Questions and Bestseller Lists
5. Accessing Gcse Titration Practise Questions Free and Paid eBooks
 - Gcse Titration Practise Questions Public Domain eBooks
 - Gcse Titration Practise Questions eBook Subscription Services

- Gcse Titration Practise Questions Budget-Friendly Options
- 6. Navigating Gcse Titration Practise Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Gcse Titration Practise Questions Compatibility with Devices
 - Gcse Titration Practise Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gcse Titration Practise Questions
 - Highlighting and Note-Taking Gcse Titration Practise Questions
 - Interactive Elements Gcse Titration Practise Questions
- 8. Staying Engaged with Gcse Titration Practise Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcse Titration Practise Questions
- 9. Balancing eBooks and Physical Books Gcse Titration Practise Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Titration Practise Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gcse Titration Practise Questions
 - Setting Reading Goals Gcse Titration Practise Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gcse Titration Practise Questions
 - Fact-Checking eBook Content of Gcse Titration Practise Questions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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