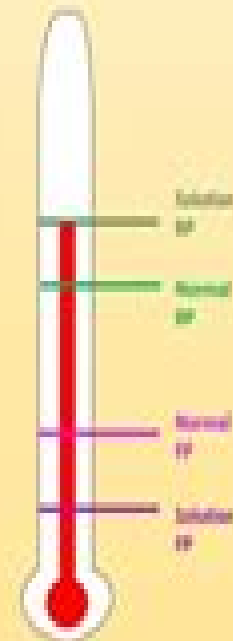
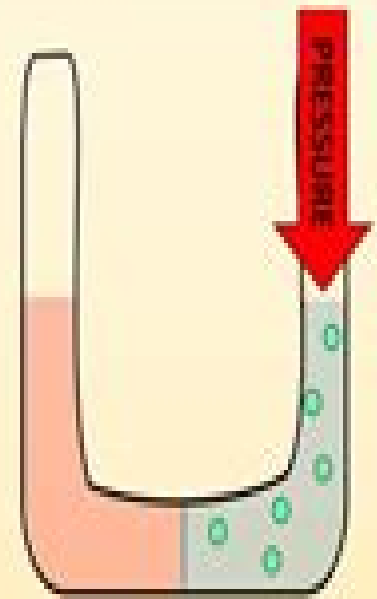


# Colligative Properties

## Objectives:

- Explain how various properties of a solution change with concentration, including boiling point, freezing point, osmotic pressure and vapor pressure.
- Quantitatively determine the degree to which these properties are affected.
  - $P_{\text{solution}} = X_{\text{solvent}} P^{\circ}_{\text{solvent}}$
  - $\Delta T_b = iK_b m$
  - $\Delta T_f = iK_f m$
  - $\Pi = iMRT$



# Lesson 16 4 Colligative Properties

**Daniel C. Tofan**



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