IN comes

Class ____ Date

7

Ionic and Metallic Bonding



BONDING AND INTERACTIONS

7.1 lons

For students using the Foundation edition, assign problems 1, 3-5, 7-12, 14, 15, 18-20.

Essential Understanding lons form when atoms gain or lose valence electrons, becoming electrically charged.

Lesson Summary

Volence Electrons. Valence electrons are the electrons in the outermost occupied energy level and are involved in ion formation.

- For a representative element, the group number equals the number of valence electrons the atom contains.
- An electron dot structure shows the symbol of the element and its valence electrons.
- Atoms tend to gain or lose the number of electrons that will provide the atom with a noble gas electron configuration.

Formation of Cations Cations are positively charged ions formed when an atom loses one or more valence electrons.

- Atoms and the cations formed from them have different properties.
- Elements in Group 1 form cations with a charge of 1+, and those in Group 2 form cations with a charge of 2+.
- Many transition metals form more than one cation and do not follow the octet rule.

Formulion of Anions Anions are negatively charged ions formed when an atom gains one or more valence electrons.

- Commonly, the name of an anion ends in -ide.
- Anions form from nonmetallic elements.
- The anions formed from halogens are known as halides.

After reading Lesson 7.1, answer the following questions.

Valence Electrons

1. What are valence electrons?

Valence electrons are the electrons in the highest occupied energy level of an

element's atoms.

2. The valence electrons largely determine the <u>chemical properties</u> of an element and are usually the only electrons used in <u>chemical bands</u>

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Kathleen Armour

Lesson19 1 Reading And Study Workbook:

Study Guide for Essentials for Nursing Practice - E-Book Patricia A. Potter, Anne G. Perry, Patricia A. Stockert, Amy Hall, Patricia Castaldi, 2018-03-15 Corresponding to the chapters in Essentials for Nursing Practice 9th Edition Study Guide for Essentials for Nursing Practice reinforces your understanding of key nursing ideas Each chapter includes a Preliminary Reading Case Study questions a Chapter Review Study Group Questions and directions to help you create your own personalized Study Chart for the chapter In addition each Chapter Review includes many different kinds of questions to keep learning the material interactive and fun Answers and rationales included on the Essentials for Nursing Practice Evolve Each chapter includes a case study with related questions allowing students to apply what they ve learned Chapter review questions include matching short answer multiple choice and true false questions provide students with evaluation and test taking practice Study group questions make it easier for students to review material with their peers site Skills performance checklists help you measure your mastery of important nursing procedures Study charts in select chapters helps provide focus and structure for students reviewing the material and key concepts NEW Content completely updated to match Nursing Essentials 9th Edition NEW Chapter on Complementary and Alternative Therapies addresses content that is now included on the NCLEX exam NEW Preliminary Readings sections provide you with an easy reference point to the related text chapter

Emergency medical services United States. National Highway Traffic Safety Administration, 1979

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