Substances and Mixtures

A. Matter: Substances and Mixtures

1. There are two main categories of matter: ______ and _____

2. Matter that is always made up of the same combination of atoms is defined as a(n) _______.

a. The two kinds of substances are elements and

b. A substance that is made up of only one kind of atom, such as oxygen, is a(n)

C. Atoms of two or more elements chemically bonded together, such as water, form (a)n

3. Atoms of two or more substances that are physically blended but are not chemically

bonded together form a(n) _______

The relative amounts of the substances that form a(n) ______ can vary.

is an example of a mixture that is a solid.

C. ______ is an example of a mixture of gases.

d. A mixture in which substances are not evenly mixed, such as granite, is called a(n) ______ mixture.

e. A mixture in which two or more substances are evenly mixed on the atomic level

but not bonded together is called a(n) _____ mixture.

f. When you look at a ______ mixture with or without a ______ all the parts look the same.

g. A homogeneous mixture can also be called a(n) ______; these terms mean the same thing.

B. How do compounds and mixtures differ?

1. The formation of a(n) ______ occurs due to a chemical ______.

2. Mixing is a(n) ______ change, so the substances that exist before mixing also exist in the mixture.

A(n) ______ has some of the same _____ properties as the substances that make up the mixture.

b. A(n) _______, in contrast, has physical properties different from the properties of the substances that make up the compound.

3. The substances that form a(n) ______ are not chemically _____ to one another.

The ______ that are combined in a mixture can be separated from one another using means.

b. In contrast, the substances that form a(n) ______ are bonded chemically, so the only way to separate them is to break the chemical ______ which is a(n) change.

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Daniel C. Tofan

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