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type of chemical reaction for this reaction we have a neutralization reaction balancing strategies here we have a
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web the balanced chemical equation for this reaction is $\text{Mg} + 2\text{HNO}_3 \text{aq} \rightarrow \text{Mg}(\text{NO}_3)_2 \text{aq} + \text{H}_2 \text{g}$ in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

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