

Name: _____

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? **They can share the markers.**

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

They can give the ones they need to each other.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** (▶) to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? _____7_____
2. Click **Pause** (⏸). Drag an electron from the left atom to the right atom. Click **Play**.
What happens? **They share an electron.**
3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.
What happens now? **They share another one.**

Gizmos Covalent Bonds Answers

Jin-Ying Zhang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Gizmos Covalent Bonds Answers

1. Understanding the eBook Gizmos Covalent Bonds Answers
 - The Rise of Digital Reading Gizmos Covalent Bonds Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmos Covalent Bonds Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Gizmos Covalent Bonds Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmos Covalent Bonds Answers
 - Personalized Recommendations
 - Gizmos Covalent Bonds Answers User Reviews and Ratings
 - Gizmos Covalent Bonds Answers and Bestseller Lists
5. Accessing Gizmos Covalent Bonds Answers Free and Paid eBooks
 - Gizmos Covalent Bonds Answers Public Domain eBooks
 - Gizmos Covalent Bonds Answers eBook Subscription Services
 - Gizmos Covalent Bonds Answers Budget-Friendly Options
6. Navigating Gizmos Covalent Bonds Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmos Covalent Bonds Answers Compatibility with Devices
 - Gizmos Covalent Bonds Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmos Covalent Bonds Answers
 - Highlighting and Note-Taking Gizmos Covalent Bonds Answers
 - Interactive Elements Gizmos Covalent Bonds Answers
8. Staying Engaged with Gizmos Covalent Bonds Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Gizmos Covalent Bonds Answers
- 9. Balancing eBooks and Physical Books Gizmos Covalent Bonds Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmos Covalent Bonds Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmos Covalent Bonds Answers
 - Setting Reading Goals Gizmos Covalent Bonds Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmos Covalent Bonds Answers
 - Fact-Checking eBook Content of Gizmos Covalent Bonds Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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