



Gimno Lab: Solubility and Temperature

Warm-up: Please Read up:

A solution generally contains at least parts of **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water, so the solution is a sugar-water solution.

Temperature affects how well solute dissolves in solution.

To begin, check that Potassium nitrate is dissolved and that Temp. of that water is 20 °C. Click OK.



1. Do these solutions, which is the solvent?

Which is the solute?

2. Click Add 10 g of potassium nitrate into the water.

3. Did all of the potassium nitrate dissolve?

4. What happened?

Activity 1:
Solubility

Read the Gizmo ready:

→ **Check points:**

- Check that the Temp. is 20 °C and that Potassium nitrate is dissolved.



Observe how the water level rises when solute is dissolved in a solution?

1. Click OK. Click Add 10 g, and observe the potassium nitrate being added into the solution. Observe how, when the 10 g of KNO₃ has completely dissolved, the water level rises. This shows that more solute can be dissolved in water than you thought. How much more?

Did all of the solute dissolve?

2. The solubility of a solute is equal to the mass of solute dissolved by the solvent at various. What is the solubility of this solute at 20 °C and 100 mL of water (up to 100 mL)?

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RS Peters

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'RS Peters' text.

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