

Name: _____ Period: _____ Date: _____

pH Gizmo

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- Acids** are substances that produce hydrogen ions (H^+) when dissolved in water. Lemon juice is an example of an acid.
 - What does lemon juice taste like? _____
 - What does it feel like if lemon juice gets in your eye? _____
- Bases** are substances that produce hydroxide ions (OH^-) when dissolved in water. Hand soap is an example of a base.
 - What does soap feel like? _____
 - What does soap taste like? _____
 - What does it feel like if soap gets in your eye? _____

Gizmo Warm-up:

The strength of an acid or base is measured on the **pH** scale. The term "pH" is short for "potential of hydrogen." It is a measure of how many excess H^+ ions there are in a solution. The pH scale runs from 0 to 14, with 0 representing the highest concentration of hydrogen ions. **Acidic** substances have a pH below 7, while **alkaline** substances (bases) have a pH above 7. Pure water has a pH of 7 and is considered **neutral**.

The pH Analysis Gizmo allows you to find the pH of a variety of liquids. In the Gizmo, check that the **Substance in the tube** is **Ammonia**, and click **Test**. Wait until the animation is finished.



- Indicators** change color in acids or bases. What is the color of the pH paper? _____
- Compare the paper to the **pH color chart**. What is the pH of ammonia? _____
- Is ammonia acidic or alkaline? _____
- Without checking, **make a prediction**: What is the pH of stomach acid? _____
- Ocean water generally has a pH of about 8. Is ocean water acidic or basic? _____
- Along with H_2O molecules and NaCl (salt), what other ions does ocean water have floating around in it? H^+ or OH^- ?
(HINT → Look back at the prior knowledge questions above.) (circle one)
- Rising global temperatures cause oceans to absorb more CO_2 from the atmosphere, which lowers the pH of the ocean water. What effect do you think this might have on ocean life?

Gizmo Ph Analysis Answers

Brendan G. Carr

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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