

Activity B:**Beta decay**Get the Gizmo ready:

- Click **Reset**, and turn off **Show equation**.
- Select **Beta decay** from the **Type of decay** menu.
- Check that **Carbon** is selected.



Question: How does beta decay change the nucleus of a radioactive atom?

1. Observe: Click **Play** and watch the animation.

A. What happens to the decaying neutron during beta decay?

The neutron turns into a proton.

B. What is emitted from the nucleus during beta decay?

Gamma ray and electron

C. What is the mass number and charge of the emitted particle?

Mass number:

0

Charge:

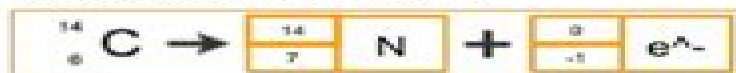
-1

2. Predict: During beta decay, a neutron is transformed into a proton and an electron (the **beta particle**), which is emitted. Gamma rays are often emitted during beta decay as well.

How will beta decay affect the atomic number and mass number of the atom?

The mass number will not change, but the atomic number will rise by one.

3. Calculate: Turn on **Write equation**. Fill in the first set of boxes with the mass number and atomic number of the daughter product and the next set of boxes with the mass number and atomic number of the beta particle. (Note: The atomic number of an electron is -1.)



Check your answer by turning on **Show equation** and clicking **Play**. Modify your equation if necessary. What isotope is produced by the beta decay of carbon-14? N-14

4. Practice: Turn off **Show equation**. Fill in the equations for the beta decay of iodine-131 and sodium-24 in the spaces below. Use the Gizmo to check your answers.



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R Sandford



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