



Name: Haley Spears

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Chemical Changes

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: acid, base, catalyst, chemical change, coefficient, conservation of matter, decomposition, dissolve, double replacement, endothermic, exothermic, indicator, ion, physical change, product, reactant, single replacement, subscript, synthesis

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)



1. A student mixes baking soda and vinegar in a glass. The results are shown at left. Do you think any new substances are being created in this mixture? If so, how do you know?

I think carbon dioxide is the material that emerged from the mixture. This is due to the fact that when vinegar and baking soda are combined, a gas is produced (carbon dioxide)

2. Suppose this was done on top of a balance. Do you think the mass would change as the reaction proceeded?

If the gas is allowed to escape, the mass will decrease rather than remain constant.

3. What do you think would happen to the mass if the reaction took place inside a sealed plastic bag?

Due to its containment, the mass would remain unchanged.

Gizmo Warm-up

A **chemical change**, (or chemical reaction) occurs when one or more substances, called **reactants**, are transformed into different substances, or **products**. In the *Chemical Changes* Gizmo, you will look for evidence of chemical changes by looking at changes you can see, touch, or smell.

To begin, check that **Reactant 1** is **Sodium** and **Reactant 2** is **Water**. Sodium is a metal so soft you can cut it with a knife.



1. Click **Play** (▶). What do you observe?

The glass of sodium was removed, and the water was added. The sodium dispersed in the water, increasing the mass.

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Joacim Rocklöv

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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