

## TYPES OF CHEMICAL BONDS

Name \_\_\_\_\_

Classify the following compounds as ionic (metal + nonmetal), covalent (nonmetal + nonmetal) or both (compound containing a polyatomic ion).

- |   |  |
|---|--|
| 1. $\text{CaCl}_2$ <u>ionic</u>         | 11. $\text{MgO}$ <u>ionic</u>              |
| 2. $\text{CO}_2$ <u>Covalent</u>        | 12. $\text{NH}_4\text{Cl}$ <u>both</u>     |
| 3. $\text{H}_2\text{O}$ <u>Covalent</u> | 13. $\text{HCl}$ <u>Covalent</u>           |
| 4. $\text{BaSO}_4$ <u>both</u>          | 14. $\text{KI}$ <u>ionic</u>               |
| 5. $\text{K}_2\text{O}$ <u>ionic</u>    | 15. $\text{NaOH}$ <u>both</u>              |
| 6. $\text{NaF}$ <u>ionic</u>            | 16. $\text{NO}_2$ <u>Covalent</u>          |
| 7. $\text{Na}_2\text{CO}_3$ <u>both</u> | 17. $\text{AlPO}_4$ <u>Both</u>            |
| 8. $\text{CH}_4$ <u>Covalent</u>        | 18. $\text{FeCl}_3$ <u>ionic</u>           |
| 9. $\text{SO}_3$ <u>Covalent</u>        | 19. $\text{P}_2\text{O}_5$ <u>Covalent</u> |
| 10. $\text{LiBr}$ <u>ionic</u>          | 20. $\text{N}_2\text{O}_5$ <u>Covalent</u> |

# Lesson 20 Covalenr Bonds Answer

**E. N. Ramsden**



## **Lesson 20 Covalent Bonds Answer:**

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