Dorte

Chemical Quantities



THE MOLE AND QUANTIFYING MATTER

10.1 The Mole: A Measurement of Matter

For students using the Foundation edition, assign problems 1–15.

Essential Understanding The mole represents a large number of very small particles.



Frayer Model The Frayer Model is a vocabulary development tool. The center of the diagram shows the concept being defined, while the quadrants around the concept are used for providing the details. Use this model when you want to understand a vocabulary term in more detail.

As you read Lesson 10.1, use the Frayer Model below. Complete the diagram for the term mole.

Definition in your own words	Facts/characteristics
A mole is 6.02 × 10 ²¹ of anything.	A mole contains Avogadro's number of anything. The mass of a mole of an element is the atomic mass of the element in grams.
How moles are used Conversion factors can be	Examples
used to change the number of particles to the number of moles or	1 mol Na = 23.0 g Na = 6.02 × 10 ²³ Na atoms.
the number of moles to the number of particles.	1 mol $H_2O = 18.0 \text{ g } H_2O$ = $6.02 \times 10^{43} H_2O$ molecules.

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Reviewing Lesson 6 Reading And Study Workbook Copyright: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "Lesson 6 Reading And Study Workbook Copyright," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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