

7 Ionic and Metallic Bonding

Big idea BONDING AND INTERACTIONS

7.1 Ions

For students using the Foundation edition, assign problems 1, 3–5, 7–12, 14, 15, 18–20.

Essential Understanding Ions form when atoms gain or lose valence electrons, becoming electrically charged.

Lesson Summary

Valence Electrons Valence electrons are the electrons in the outermost occupied energy level and are involved in ion formation.

- ▶ For a representative element, the group number equals the number of valence electrons the atom contains.
- ▶ An electron dot structure shows the symbol of the element and its valence electrons.
- ▶ Atoms tend to gain or lose the number of electrons that will provide the atom with a noble gas electron configuration.

Formation of Cations Cations are positively charged ions formed when an atom loses one or more valence electrons.

- ▶ Atoms and the cations formed from them have different properties.
- ▶ Elements in Group 1 form cations with a charge of $1+$, and those in Group 2 form cations with a charge of $2+$.
- ▶ Many transition metals form more than one cation and do not follow the octet rule.

Formation of Anions Anions are negatively charged ions formed when an atom gains one or more valence electrons.

- ▶ Commonly, the name of an anion ends in *-ide*.
- ▶ Anions form from nonmetallic elements.
- ▶ The anions formed from halogens are known as halides.

After reading Lesson 7.1, answer the following questions.

Valence Electrons

1. What are valence electrons?

Valence electrons are the electrons in the highest occupied energy level of an element's atoms.

2. The valence electrons largely determine the *chemical properties* of an element and are usually the only electrons used in *chemical bonds*.

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Reading Instruction in the Armed Forces Lowry Waring Harding, James Burgett Burr, 1950 **A Workbook for Reading Argumentative Texts** James E. Scheuermann, 2022-07-26 This Workbook is a collection of exercises and case studies designed to serve as a companion to Reading Argumentative Texts Analytic Tools to Improve Understanding The exercises and case studies track each of the chapters of Reading and provide opportunities for students to hone their skills at using the analytic tools presented in Reading and to acquire additional analytic tools and concepts These tools are illustrated through the analysis of complete essays from the mass media speeches a sermon and passages from academic works The approach is flexible and practical and avoids academic jargon and specific theories of argumentation As is the case with Reading this Workbook is grounded in two principles First that the meaning of an argumentative text is to be found in the statements that constitute the argument itself in other statements that are more or less directly related to the argument and in the structure and context of the text Accordingly while this book discusses the analysis of arguments argument types and errors in argumentation fallacies it focuses equally on the other sources of meaning of a text Second there is no single authoritative reading of an argumentative text The interplay of these two premises informs the view that analyzing and understanding an argumentative text is an art and that within certain well defined parameters there are better and worse readings of a text and not right or wrong readings The principal sources of meaning discussed include 1 the structure of the text and so the book examines six types of introductions and teaches how to outline and summarize 2 key sentences phrases and words in a text so the book discusses ambiguity the difference between factual and normative statements irony and rhetoric 3 context intellectual social political cultural and physical context and 4 the logical connections between terms in an argument including the four different types of arguments fallacies and the distinction between necessary and sufficient conditions The book is designed to be used in late high school or early college critical reading critical thinking rhetoric or writing courses **Waves, Sound, and Light** ,2005

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