

Chapter 9 Carbon Chemistry

Section 9.2 Substituted Hydrocarbons**(pages 272–274)**

This section discusses organic compounds that contain atoms of elements other than carbon and hydrogen. It also explains the relationship between the properties of organic compounds and functional groups.

Reading Strategy (page 272)

Monitoring Your Understanding As you read, complete the table by connecting each functional group with the type of compound that contains the functional group. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Connecting Functional Groups to Types of Compounds	
Functional Group	Type of Compound
–OH	
–COOH	
–NH ₂	

- Name the two main products when methane and chlorine react.
 - _____
 - _____
- To which environmental problem have researchers connected halocarbons containing chlorine and fluorine? _____

- Describe a substituted hydrocarbon. _____

- Is the following sentence true or false? The functional group in a substituted hydrocarbon determines the properties of the compound. _____

Alcohols (page 273)

- Methanol and ethanol are two examples of a class of organic compounds called _____.
- The functional group in an alcohol is represented as –OH and is called a(n) _____ group.
- Identify two ways a halocarbon can be produced.
 - _____
 - _____

Hydrocarbon Compounds Section Review Answer Key

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Table of Contents Hydrocarbon Compounds Section Review Answer Key

1. Understanding the eBook Hydrocarbon Compounds Section Review Answer Key
 - The Rise of Digital Reading Hydrocarbon Compounds Section Review Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Hydrocarbon Compounds Section Review Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hydrocarbon Compounds Section Review Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hydrocarbon Compounds Section Review Answer Key
 - Personalized Recommendations
 - Hydrocarbon Compounds Section Review Answer Key User Reviews and Ratings

- Hydrocarbon Compounds Section Review Answer Key and Bestseller Lists
- 5. Accessing Hydrocarbon Compounds Section Review Answer Key Free and Paid eBooks
 - Hydrocarbon Compounds Section Review Answer Key Public Domain eBooks
 - Hydrocarbon Compounds Section Review Answer Key eBook Subscription Services
 - Hydrocarbon Compounds Section Review Answer Key Budget-Friendly Options
- 6. Navigating Hydrocarbon Compounds Section Review Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Hydrocarbon Compounds Section Review Answer Key Compatibility with Devices
 - Hydrocarbon Compounds Section Review Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hydrocarbon Compounds Section Review Answer Key
 - Highlighting and Note-Taking Hydrocarbon Compounds Section Review Answer Key
 - Interactive Elements Hydrocarbon Compounds Section Review Answer Key
- 8. Staying Engaged with Hydrocarbon Compounds Section Review Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hydrocarbon Compounds Section Review Answer Key
- 9. Balancing eBooks and Physical Books Hydrocarbon Compounds Section Review Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hydrocarbon Compounds Section Review Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hydrocarbon Compounds Section Review Answer Key
 - Setting Reading Goals Hydrocarbon Compounds Section Review Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hydrocarbon Compounds Section Review Answer Key
 - Fact-Checking eBook Content of Hydrocarbon Compounds Section Review Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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