

Hunter



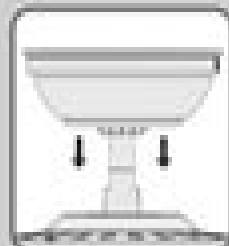
Universal 3 Speed Ceiling Fan Remote Model 99122, 99123

If you are installing multiple remote-controlled fans on the same circuit breaker, you may need to perform a few extra steps to prevent interference or faulty operation of your remote controls. Go to www.hunterfan.com/RACs and click "How do I properly install multiple remote-controlled fans?" for more information.

Installation



Before installing the Universal Remote Control Receiver, use the remote control to set the fan speed to "off" and the light to "off" before proceeding with installation.



Insert the receiver into the fan's mounting bracket. Be sure to align the receiver's pins with the bracket's slots. After wiring is complete, it may be helpful to rotate the receiver clockwise or counter-clockwise to align the pins with the slots.



OR



OR



OR

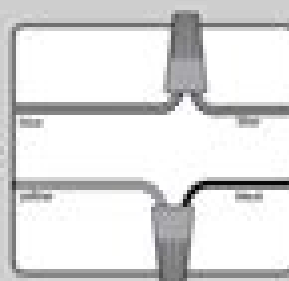


Choose the hanging system that most closely resembles the one used by your fan and install the receiver as shown and wire as directed.

Note: Some fans may have a separate power switch for the fan and light. If so, the receiver should be installed in the fan's power switch. The receiver and fan must remain properly grounded.

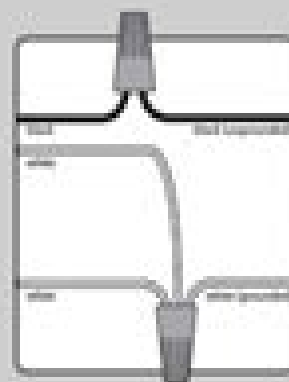
The receiver should be installed in a location that is easily accessible for maintenance and is protected from moisture.

Wiring



CONNECT WIRE FROM RECEIVER TO FAN: Using the wiring as shown, connect the wires from the receiver to the fan. The fan's power supply wires with white stripes and blue stripes. Connect the wires from the receiver to the fan's power supply wires.

If you're not sure about wire colors or connections, please consult a qualified electrician.



CONNECT WIRE FROM RECEIVER AND FAN TO WIRE FROM OUTLET BOX: Using the wiring as shown, connect the wires from the fan's power supply to the outlet box. The fan's power supply wires with white stripes and blue stripes. Connect the wires from the outlet box to the fan's power supply wires.

AFTER ALL WIRES ARE CONNECTED: Double-check all wire connections to be sure they are correct.

For more information on how to install the Hunter Universal 3 Speed Ceiling Fan Remote, please visit our website at www.hunterfan.com. For more information on how to install the Hunter Universal 3 Speed Ceiling Fan Remote, please visit our website at www.hunterfan.com.

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Hunter R61operating Manual

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web k sp of calcium hydroxide 1 author j m mccormick last update ocober 13 2013 introduction an equilibrium constant K is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of K

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web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the ksp of the compound you will do this by titrating the prepared ca oh 2 solution with a standard hydrochloric acid solution

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of ca oh is represented in equation form as shown below ca oh s ca aq 2oh aq the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is ca oh 2 ca2 2 oh

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oh 2 m ca in saturated ca oh 2 m ksp of ca oh 2 average ksp of ca oh 2

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web the chemical equation used was $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ the solubility product was determined to be $K_{sp} \text{Ca(OH)}_2$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as $\text{Ca}_3(\text{PO}_4)_2$ is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K_{sp} \text{Ca}_3(\text{PO}_4)_2$

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web lab report 8 determinin of calcium hydroxide the purpose of this was to titrat e a saturated ca oh 2 soultion with a standard hcl solution and find the ksp or the oh concentration was f ound and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochlo

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web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of ca oh 2 is represented in equation form lution of ca on s nronic solid that is sparingly soluble in water a saturated aqueous as shown below

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for ca was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the ksp of calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8 1 2 x 11 sheet of paper

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the ksp for calcium hydroxide Ca^{2+} OH^- 2.0×10^{-2} ksp 5×10^{-4} find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is 5×10^{-6}

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