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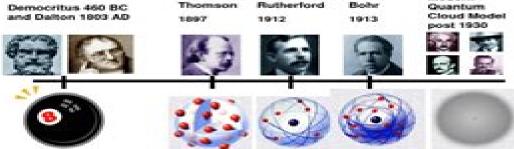
## History of the Atom Timeline

Modern.

Dalton's

Atomic Theory





#### John Dalton (1766 - 1844):

John Dalton was an English chemist. His ideas form the atomic theory of matter. Here are his ideas.

- All elements are composed (made up) of atoms. It is impossible to divide or destroy an atom.
- All atoms of the same elements are alike. (One atom of oxygen is like another atom of oxygen.)
- Atoms of different elements are different. (An atom of oxygen is different from an atom of hydrogen.)
- Atoms of different elements combine to form a compound. These atoms have to be in definite whole number ratios.
   For example, water is a compound made up of 2 atoms of hydrogen and 1 atom of oxygen (a ratio of 2:1). Three atoms of hydrogen and 2 atoms of oxygen cannot combine to make water.
- 1. What is the name of John Dalton's theory? THE ATOMIC THEORY OF MATTER
- 2. What are elements made of? ELEMENTS ARE MADE OF ATOMS
- 3. An atom of hydrogen and an atom of carbon are DIFFERENT
- 4. What are compounds made of? COMPOUNDS ARE MADE OF ATOMS OF TWO OR MORE DIFFERENT ELEMENTS:
- 5. The ratio of atoms in HCl is: a) 1:3 b) 2:1 c) 1:1
- 6. The ratio of atoms in H<sub>2</sub>O is: a) 1:3 b) 2:1 c) 1:1 2 atoms of hydrogen and 1 atom of oxygen

# **History Of The Atom Answersheet**

**JS Bruner** 

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