

Name: Carter David Date: _____

Student Exploration: Isotopes

Vocabulary: atomic number, band of stability, half-life, isotope, isotope notation, mass number, radioactive, radioisotope

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. What particles make up an atom? protons, neutrons, and electrons
2. Which of these are found in the nucleus? Protons and Neutrons
3. Which particles are charged? Electrons

Gizmo Warm-up

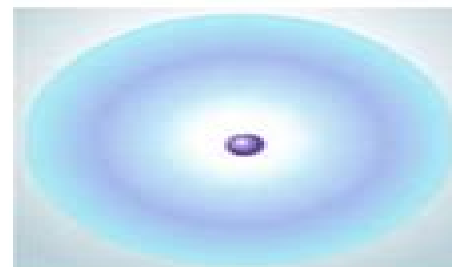
You may think that all atoms of an element are the same, but that is not the case. Atoms of an element can come in several different versions. Some of these versions are stable, while others can break down. In the *Isotopes* Gizmo, you will explore different versions of the elements.

To begin, check that **Hydrogen-1** is selected. Check that **Show isotope notation** is selected.

1. Click the up and down arrows for protons and neutrons and notice what changes.
 - A. Which particle determines the element? The number of protons
 - B. Which particle creates a different version of the same element? The number of neutrons

Atoms of the same element with different numbers of neutrons are called **isotopes**.

2. Set protons and neutrons to 6.
 - A. Which isotope was created? Carbon-12
 - B. What percentage of the element consists of this isotope? 98.93%
 - C. Add a neutron. Which isotope do you have now? Carbon-13



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Katrin Zwirglmaier



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Table of Contents Gizmo Student Exploratin Half Life Activity A

1. Understanding the eBook Gizmo Student Exploratin Half Life Activity A
 - The Rise of Digital Reading Gizmo Student Exploratin Half Life Activity A
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Student Exploratin Half Life Activity A
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Student Exploratin Half Life Activity A
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Student Exploratin Half Life Activity A
 - Personalized Recommendations
 - Gizmo Student Exploratin Half Life Activity A User Reviews and Ratings
 - Gizmo Student Exploratin Half Life Activity A and Bestseller Lists
5. Accessing Gizmo Student Exploratin Half Life Activity A Free and Paid eBooks

- Gizmo Student Exploratin Half Life Activity A Public Domain eBooks
 - Gizmo Student Exploratin Half Life Activity A eBook Subscription Services
 - Gizmo Student Exploratin Half Life Activity A Budget-Friendly Options
6. Navigating Gizmo Student Exploratin Half Life Activity A eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Student Exploratin Half Life Activity A Compatibility with Devices
 - Gizmo Student Exploratin Half Life Activity A Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Student Exploratin Half Life Activity A
 - Highlighting and Note-Taking Gizmo Student Exploratin Half Life Activity A
 - Interactive Elements Gizmo Student Exploratin Half Life Activity A
 8. Staying Engaged with Gizmo Student Exploratin Half Life Activity A
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Student Exploratin Half Life Activity A
 9. Balancing eBooks and Physical Books Gizmo Student Exploratin Half Life Activity A
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Student Exploratin Half Life Activity A
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Gizmo Student Exploratin Half Life Activity A
 - Setting Reading Goals Gizmo Student Exploratin Half Life Activity A
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Gizmo Student Exploratin Half Life Activity A
 - Fact-Checking eBook Content of Gizmo Student Exploratin Half Life Activity A
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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