

4. **Predict:** What do you think will happen if you fire a 7-eV photon at the atom of **Gas A**? How about a 13-eV or a 19-eV photon?

The electron will move to orbitals that are farther away.

5. **Gather data:** Test your predictions with the Gizmo and fill in the table below. (The first row has been filled in for you.)

Photon energy	Effect on electron	Energy of emitted photon(s)
4 eV	Electron moves up to $n = 2$ and then back down to $n = 1$.	4 eV
7 eV	The electron climbs to $n=3$ before descending to $n=1$.	4 eV and 3 eV
13 eV	The electron climbs to $n=4$ before descending to $n=1$.	4 eV and 9 eV
19 eV	The electron climbs to $n=5$ before descending to $n=1$.	4 eV and 15 eV

6. **Analyze:** Find the total energy of each set of emitted photons. How does each sum relate to the energy of the absorbed photon?

Yes, it has only been divided.

7. **Explore:** With the **Energy (eV)** set to 19 eV, click **Fire** six times. Record the energy of the emitted photons each time. Record the results of each trial below.

Trial	Energy of emitted photons	Trial	Energy of emitted photons
1	3 eV, 4 eV, 6 eV and 6 eV	4	3 eV, 4 eV, 6 eV and 6 eV
2	6 eV and 13 eV	5	3 eV, 4eV, 12 eV
3	3 eV, 4 eV, 6 eV and 6 eV	6	4 eV and 15 eV

8. **Analyze:** When an electron moves from a higher orbit to a lower one, does it always follow the same path? Explain.

No, it has multiple ways to distribute the energy used to move it to the higher orbital.

Gizmo Bohr Model Of Hydrogen Answers Key

Yicheng Fang



Gizmo Bohr Model Of Hydrogen Answers Key:

Unveiling the Power of Verbal Artistry: An Emotional Sojourn through **Gizmo Bohr Model Of Hydrogen Answers Key**

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Table of Contents Gizmo Bohr Model Of Hydrogen Answers Key

1. Understanding the eBook Gizmo Bohr Model Of Hydrogen Answers Key
 - The Rise of Digital Reading Gizmo Bohr Model Of Hydrogen Answers Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Bohr Model Of Hydrogen Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Bohr Model Of Hydrogen Answers Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Bohr Model Of Hydrogen Answers Key
 - Personalized Recommendations

- Gizmo Bohr Model Of Hydrogen Answers Key User Reviews and Ratings
- Gizmo Bohr Model Of Hydrogen Answers Key and Bestseller Lists
- 5. Accessing Gizmo Bohr Model Of Hydrogen Answers Key Free and Paid eBooks
 - Gizmo Bohr Model Of Hydrogen Answers Key Public Domain eBooks
 - Gizmo Bohr Model Of Hydrogen Answers Key eBook Subscription Services
 - Gizmo Bohr Model Of Hydrogen Answers Key Budget-Friendly Options
- 6. Navigating Gizmo Bohr Model Of Hydrogen Answers Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Bohr Model Of Hydrogen Answers Key Compatibility with Devices
 - Gizmo Bohr Model Of Hydrogen Answers Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Bohr Model Of Hydrogen Answers Key
 - Highlighting and Note-Taking Gizmo Bohr Model Of Hydrogen Answers Key
 - Interactive Elements Gizmo Bohr Model Of Hydrogen Answers Key
- 8. Staying Engaged with Gizmo Bohr Model Of Hydrogen Answers Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Bohr Model Of Hydrogen Answers Key
- 9. Balancing eBooks and Physical Books Gizmo Bohr Model Of Hydrogen Answers Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Bohr Model Of Hydrogen Answers Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Bohr Model Of Hydrogen Answers Key
 - Setting Reading Goals Gizmo Bohr Model Of Hydrogen Answers Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Bohr Model Of Hydrogen Answers Key
 - Fact-Checking eBook Content of Gizmo Bohr Model Of Hydrogen Answers Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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