

### Activity A:

#### Subatomic particles

#### Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



**Question:** What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** ( $A$ ), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** ( $Z$ ).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons ( $N$ ) in an atom?
- C. Which particle (proton, neutron, or electron) has a  
Positive charge?  
Negative charge?

neutrons

by the number of protons

electron

proton

Electrical charge?

3. **Observe:** Turn on **Electron shells**. Electrons are arranged in layers called shells. The number of shells is equal to the atomic number ( $Z$ ). The number of electrons in each shell is also equal to the atomic number ( $Z$ ). The number of protons in each shell is also equal to the atomic number ( $Z$ ). The number of neutrons in each shell is also equal to the atomic number ( $Z$ ).

- A. What are the shells called?
- B. What are the shells called?
- C. What are the shells called?

Electron shells

Electron shells

Electron shells

4. **Observe:** Turn on **Electron shells**. Answer the following questions.

- A. How many electrons are in a neutral atom of Helium?
- B. How many electrons are in a neutral atom of Hydrogen?
- C. What is the mass number of an atom with 2 protons and 2 neutrons?
- D. What is the mass number of an atom with 2 protons and 1 neutron?
- E. How many electrons are in Helium?

2

1

4

3

2

# Gizmo Answers Element Builder

**Yicheng Fang**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name Yicheng Fang.

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## **Table of Contents Gizmo Answers Element Builder**

1. Understanding the eBook Gizmo Answers Element Builder
  - The Rise of Digital Reading Gizmo Answers Element Builder
  - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answers Element Builder
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gizmo Answers Element Builder
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Answers Element Builder
  - Personalized Recommendations
  - Gizmo Answers Element Builder User Reviews and Ratings
  - Gizmo Answers Element Builder and Bestseller Lists

5. Accessing Gizmo Answers Element Builder Free and Paid eBooks
  - Gizmo Answers Element Builder Public Domain eBooks
  - Gizmo Answers Element Builder eBook Subscription Services
  - Gizmo Answers Element Builder Budget-Friendly Options
6. Navigating Gizmo Answers Element Builder eBook Formats
  - ePub, PDF, MOBI, and More
  - Gizmo Answers Element Builder Compatibility with Devices
  - Gizmo Answers Element Builder Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Gizmo Answers Element Builder
  - Highlighting and Note-Taking Gizmo Answers Element Builder
  - Interactive Elements Gizmo Answers Element Builder
8. Staying Engaged with Gizmo Answers Element Builder
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gizmo Answers Element Builder
9. Balancing eBooks and Physical Books Gizmo Answers Element Builder
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gizmo Answers Element Builder
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Answers Element Builder
  - Setting Reading Goals Gizmo Answers Element Builder
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gizmo Answers Element Builder
  - Fact-Checking eBook Content of Gizmo Answers Element Builder
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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