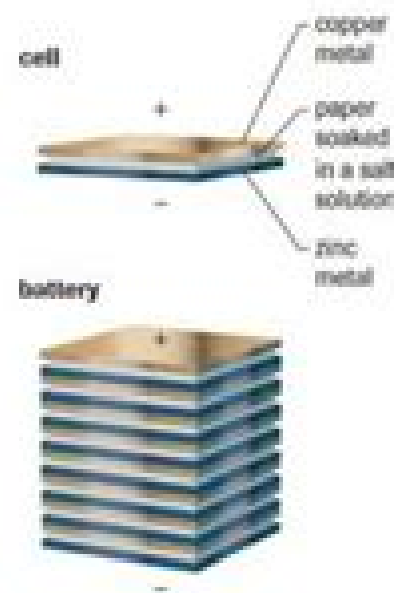


Introduction to Electrochemistry

- ▶ An *electric cell* converts chemical energy into electrical energy
 - ▶ Alessandro Volta invented the first electric cell but got his inspiration from Luigi Galvani. Galvani's crucial observation was that two different metals could make the muscles of a frog's legs twitch. Unfortunately, Galvani thought this was due to some mysterious "animal electricity". It was Volta who recognized this experiment's potential.
 - ▶ An electric cell produces very little electricity, so Volta came up with a better design:
- ▶ A *battery* is defined as two or more electric cells connected in series to produce a steady flow of current
 - ▶ Volta's first battery consisted of several bowls of brine ($\text{NaCl}_{(\text{aq})}$) connected by metals that dipped from one bowl to another
 - ▶ His revised design, consisted of a sandwich of two metals separated by paper soaked in salt water.



Experiment 22 Electrochemical Cells Post Lab Answers

Sasha Hu



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Scientific and Technical Aerospace Reports ,1990 *Energy Research Abstracts* ,1990 NASA Technical Memorandum ,1963 Index Medicus ,2001-06 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings

Cumulated Index Medicus ,1989 Nuclear Science Abstracts ,1973 *Australian Journal of Experimental Agriculture* ,2005 **The Gating and Maintenance of sleep and Wake: New Circuits and Insights** Michael

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