Study guide: Energy (Ch 3 and 4)

- What is Free Energy? (ΔG)
- What does it tell you?
- 3. How does it relate to entropy?
- What is Std. Free Energy? (ΔG°)
- 5. How does ΔG° relate to the equilibrium constant and what is equilibrium?
- 6. What is the equation that relates ΔG to ΔG* and the concentrations of reactants and products?
- How are reactions with a +ΔG coupled to reactions with a -ΔG?
- 8. How can you tell if a reaction with a +ΔG can be coupled to a reaction with a -ΔG?
- 9. What do enzymes do?
- 10. What are high energy reaction intermediates?
- 11. What are activated carriers? (know the ones we discussed)
- 12. What are anabolism and catabolism?
- Glycolysis, Citric acid cycle, Electron transport (chemiosmotic coupling and oxidative phosphorylation)
 - -What does each process accomplish?
 - -What are the starting molecules and end products of each?
 - -Where in the cell does each process occur?
 Eukaryotes?
 Prokaryotes?
 - -How are activated carriers generated and used? (know examples we discussed)
 - -What is fermentation?

Energy In A Cell Study Guide

Thomas Griffiths

Energy In A Cell Study Guide:

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