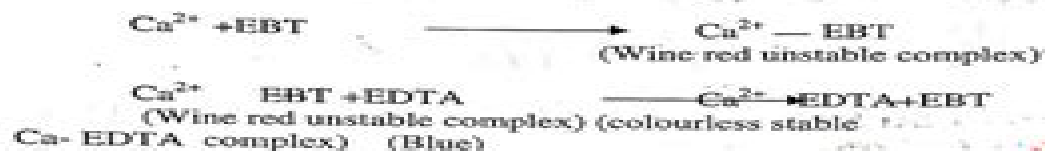
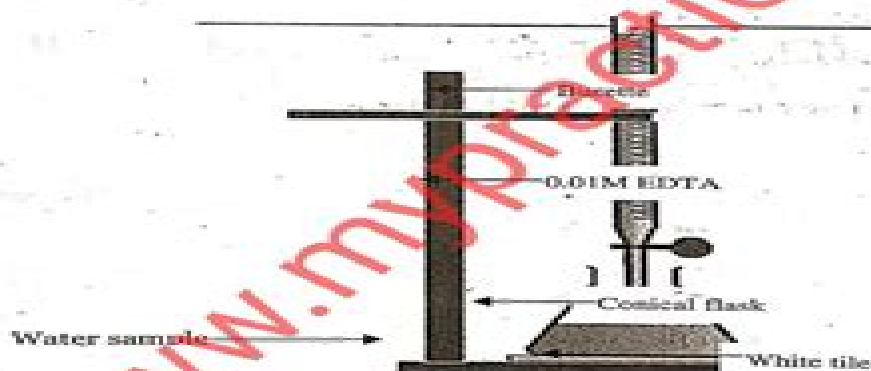


alkaline earth metal ions  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ . The EDTA reagent can be used to measure the total quantity of dissolved  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  ions in a water sample. Determination of hardness of water is based on the fact that when indicator, Eriochrome Black -T (EBT) is added to hard water in alkaline medium, it forms wine red colour complex with  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  ions.



Thus the total hardness of a water sample can be estimated by titration with standard solution of EDTA.

### VIII Practical set-up / Circuit diagram / Work Situation



### IX Resources Required

S. No.	Name of Resource	Suggested Broad Specification	Quantity
1.	Burette	Borosil glass, Capacity 25 ml / 50 ml	One per group
2.	Conical flask	Borosil glass, Capacity 250 ml/100 ml	One per group
3.	Beaker	Borosil glass, Capacity 250 ml/100 ml	One per group
4.	Pipette	Borosil glass, Capacity 25ml / 10 ml,	One per group
5.	EDTA solution	0.01 M	One per group
6.	Buffer solution	pH = 10	As per Requirement
7.	Eriochrome Black T indicator		
8.	Water sample	Hard water sample	

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