

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

সংকেত, $m = 20$ g

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে উল্লিখিত N -পদ্ধতিতে 10 mg এর মধ্যে কয় পরিমাণ দ্রব V -পদ্ধতিতে 0.2 g 2% $\left(\frac{W}{V}\right)$ HCl দ্রব দিতে কাজ করা হয়েছে।

যদিও N -পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও V -পদ্ধতিটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশন ও বিশুদ্ধতা বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2 g কঠিন বা $2-4 \text{ mL}$ তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2S এর পরিবর্তে ক্লোরোসালফোনিক ব্যবহার করা হয় ফলে পরিবেশ দূষণ ঘটে না।



অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

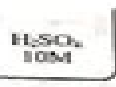
উদাহরণ



পাত্র-ক



পাত্র-খ



পাত্র-ঘ

কি. কে. ১০। এর ক. ১/

১. উদাহার দ্রব কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঙ্গার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহারের অংশের দ্রব থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1 M ঘনমাত্রার দ্রব তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪

৪ নং প্রশ্নের উত্তর

১. বিশ্লেষণের নিমিত্ত বীমের উপর রাইডার স্থাপন করলে বীমের প্রতি সারকে অন্য যে দর পাওয়া যায়, তাকে রাইডার দ্রবক বলে।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশন ও বিশুদ্ধতা বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2 g কঠিন বা $2-4 \text{ mL}$ তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে। অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

৩. উদাহারের দ্রবপট্ট হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M দ্রব। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M দ্রব যখন পাত্র H_2SO_4 দ্রব। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঙ্গার প্রতীক অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঙ্গার প্রতীক নিম্নে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আবশ্য	ক্ষয়কারী রাসায়নিক উপাদান কলতে যে ক্ষয়জনক রাসায়নিক দ্রব্যকে খুবই দ্রুত ক্ষয় সম্পর্কে ত্বক ও অন্যান্য পদার্থের ক্ষয়জনক ক্ষতি সঞ্চিত হয়। এতে ত্বক পুড়ে যায় এবং অন্যান্য পদার্থ গলে যায়।
উদাহরণ	পাত্র কঠিন এসিডসমূহ (HCl , HNO_3 , H_2SO_4), ক্ষয়কারী (NaOH , KOH), H_2O_2 , অক্সালিক অ্যাসিড, ক্লোরিন, মূল্য ইত্যাদি।
সংরক্ষণ পদ্ধতি	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র সংরক্ষণে, এরপর পাত্রের পিছনকে পোতলে রাখতে হবে।

উদাহারের দ্রবপট্ট হলো 10 M H_2SO_4 দ্রব এবং এর থেকে 1 M ঘনমাত্রার দ্রব প্রস্তুত করতে হবে 1 mL ও এর কাছাকাছি ব্যবহার করতে হবে।

1 M কাছাকাছি হলো 100 mL আয়তনমিতিক দ্রব ও এর টি হলো পিপিটি। সেমি-মাইক্রো আয়তনমিতিক দ্রবের আয়তন 100 mL । তাই আমাদের 100 mL 1 M ঘনমাত্রার H_2SO_4 দ্রব তৈরি করতে হবে। দ্রবের অনুকরণ দ্রব হতে আমরা জানি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10 mL H_2SO_4 দ্রবের সঙ্গে পানি যোগ করতে হবে $(100 - 10) \text{ mL} = 90 \text{ mL}$ । অতএব, পিপিটির সাহায্যে 10 mL দ্রব ঘনমাত্রার দ্রব নিজে তার সাথে আয়তনমিতিক দ্রবকে 90 mL পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার দ্রব পাওয়া যাবে।

উদাহরণ



চিত্র-A



চিত্র-B



চিত্র-C

কি. কে. ১০। এর ক. ১/

১. অক্সিডাইজিং এজেন্ট কাকে বলে?
২. সাদা দ্রবক $\text{K}_2\text{Cr}_2\text{O}_7$ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A ডিফের হ্যাঙ্গারের উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C ডিফের সাথে সম্পর্কিত রাসায়নিক দ্রব্যের ব্যবহারকালে খান্দা কৃতি দ্রব কবীর উপর সম্পর্কে বিশ্লেষণ করে। ৪

৪ নং প্রশ্নের উত্তর

১. অক্সিডাইজিং এজেন্ট এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চেপক যা দ্রব বা উপাদানসমূহকে জারণ প্রক্রিয়ায় আনতে সক্ষম করে বাবার সঞ্চেপ করে।

২. সামান্য $\text{K}_2\text{Cr}_2\text{O}_7$ এর ঘন কখনো অসীম বা শূন্য হতে পারে না। উল্লিখিত ব্যাখ্যা করার জন্য নিম্নের উত্তরসূরী মাইক্রোপট বিশ্লেষণ করি-

$$A + B \longrightarrow C + D$$

Hsc Chemistry 1 First Chapter Notes

Marcel A. Müller



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