

TYPES OF CHEMICAL BONDS

Name _____

Classify the following compounds as ionic (metal + nonmetal), covalent (nonmetal + nonmetal) or both (compound containing a polyatomic ion).

- | | | | |
|-----------------------------|-----------------|----------------------------|-----------------|
| 1. CaCl_2 | <u>ionic</u> | 11. MgO | <u>ionic</u> |
| 2. CO_2 | <u>Covalent</u> | 12. NH_4Cl | <u>both</u> |
| 3. H_2O | <u>Covalent</u> | 13. HCl | <u>Covalent</u> |
| 4. BaSO_4 | <u>both</u> | 14. KI | <u>ionic</u> |
| 5. K_2O | <u>ionic</u> | 15. NaOH | <u>both</u> |
| 6. NaF | <u>ionic</u> | 16. NO_2 | <u>Covalent</u> |
| 7. Na_2CO_3 | <u>both</u> | 17. AlPO_4 | <u>Both</u> |
| 8. CH_4 | <u>Covalent</u> | 18. FeCl_3 | <u>ionic</u> |
| 9. SO_3 | <u>Covalent</u> | 19. P_2O_5 | <u>Covalent</u> |
| 10. LiBr | <u>ionic</u> | 20. N_2O_5 | <u>Covalent</u> |

Holt Chemistry Covalent Compounds Review Answers

M Mark



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Table of Contents Holt Chemistry Covalent Compunds Review Answers

1. Understanding the eBook Holt Chemistry Covalent Compunds Review Answers
 - The Rise of Digital Reading Holt Chemistry Covalent Compunds Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Covalent Compunds Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Covalent Compunds Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Covalent Compunds Review Answers
 - Personalized Recommendations
 - Holt Chemistry Covalent Compunds Review Answers User Reviews and Ratings
 - Holt Chemistry Covalent Compunds Review Answers and Bestseller Lists
5. Accessing Holt Chemistry Covalent Compunds Review Answers Free and Paid eBooks
 - Holt Chemistry Covalent Compunds Review Answers Public Domain eBooks
 - Holt Chemistry Covalent Compunds Review Answers eBook Subscription Services
 - Holt Chemistry Covalent Compunds Review Answers Budget-Friendly Options
6. Navigating Holt Chemistry Covalent Compunds Review Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Holt Chemistry Covalent Compunds Review Answers Compatibility with Devices
 - Holt Chemistry Covalent Compunds Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Holt Chemistry Covalent Compunds Review Answers
 - Highlighting and Note-Taking Holt Chemistry Covalent Compunds Review Answers
 - Interactive Elements Holt Chemistry Covalent Compunds Review Answers
8. Staying Engaged with Holt Chemistry Covalent Compunds Review Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Covalent Compunds Review Answers
9. Balancing eBooks and Physical Books Holt Chemistry Covalent Compunds Review Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Covalent Compunds Review Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Holt Chemistry Covalent Compunds Review Answers
- Setting Reading Goals Holt Chemistry Covalent Compunds Review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Chemistry Covalent Compunds Review Answers
- Fact-Checking eBook Content of Holt Chemistry Covalent Compunds Review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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