

LONDON COLLEGE UNIVERSITY CHEMICAL – 4000 (004)

NAME: _____

ENTRY NUMBER: _____

ICM: 1000 (CONCEPTS IN ORGANIC CHEMISTRY)

ICM 10

Course : ICM 1000 (004)

Days: (1st- _____)

Hours : 1:00 Minutes

Time : 10:00 - 11:00

PART – A

10 x 10 = 100 marks

Answer ALL the questions.

1. State three bonding/intermolecular principles.
2. Describe hydrogen bonding in alcohols in the context of their boiling points.
3. Write the chemical formulae for $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$ and $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$.
4. Describe the structure of hydrogen bonding in H_2O molecules.
5. What intermolecular forces between molecules govern the boiling point of an alcohol and the rate of evaporation of an alcohol in comparison to the boiling point of an ether containing the same number of carbons?
6. Describe and explain how liquid films emerge from a liquid when the surface tension is low.
7. What are alcohols – definition and/or?
8. What are alcohols? Give an example.
9. Suggest a test for alcohols (A.E.C., 1999), (4g)
10. What are organic solvents? Give 2 examples.

PART – B

10 x 10 = 100 marks

Answer any FIVE questions.

11. What are solid and liquid state changes? What intermolecular forces are present in these states?
12. Explain the role of intermolecular forces in the boiling point of a liquid.
13. Explain the role of intermolecular forces in the boiling point of a liquid.
14. Explain the role of intermolecular forces in the boiling point of a liquid.
15. Explain the role of intermolecular forces in the boiling point of a liquid.
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19. Explain the role of intermolecular forces in the boiling point of a liquid.
20. Explain the role of intermolecular forces in the boiling point of a liquid.
21. Explain the role of intermolecular forces in the boiling point of a liquid.
22. Explain the role of intermolecular forces in the boiling point of a liquid.

PART – C

10 x 10 = 100 marks

Answer any FIVE questions.

23. What are organic solvents? Explain how they are used. Describe the properties of organic solvents.
24. Explain the role of intermolecular forces in the boiling point of a liquid.
25. Explain the role of intermolecular forces in the boiling point of a liquid.
26. Explain the role of intermolecular forces in the boiling point of a liquid.
27. Explain the role of intermolecular forces in the boiling point of a liquid.
28. Explain the role of intermolecular forces in the boiling point of a liquid.
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31. Explain the role of intermolecular forces in the boiling point of a liquid.
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