

Solution Manual for Chemistry, 13th Edition, Raymond Chang, Jason Overby

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CHAPTER 1 CHEMISTRY: THE STUDY OF CHANGE

Problem Categories

Biological: 1.26, 1.50, 1.71, 1.72, 1.80, 1.86, 1.96, 1.97, 1.105, 1.114.

Conceptual: 1.3, 1.4, 1.17, 1.18, 1.11, 1.12, 1.56, 1.64, 1.99, 1.94, 1.101, 1.103, 1.117.

Environmental: 1.72, 1.89, 1.91, 1.98, 1.109, 1.112.

Industrial: 1.53, 1.57, 1.83.

Difficulty Level

Easy: 1.3, 1.9, 1.10, 1.11, 1.12, 1.17, 1.23, 1.24, 1.25, 1.26, 1.27, 1.28, 1.31, 1.32, 1.33, 1.34, 1.35, 1.36, 1.56, 1.57, 1.66, 1.79, 1.82, 1.86, 1.91.

Medium: 1.4, 1.18, 1.17, 1.38, 1.39, 1.40, 1.41, 1.42, 1.43, 1.44, 1.45, 1.46, 1.47, 1.48, 1.49, 1.50, 1.51, 1.52, 1.53, 1.54, 1.55, 1.58, 1.59, 1.60, 1.61, 1.62, 1.63, 1.64, 1.65, 1.72, 1.73, 1.74, 1.75, 1.76, 1.77, 1.78, 1.80, 1.81, 1.83, 1.84, 1.85, 1.87, 1.93, 1.96, 1.97, 1.98.

Difficult: 1.67, 1.68, 1.69, 1.70, 1.71, 1.88, 1.89, 1.90, 1.92, 1.94, 1.95, 1.99, 1.100, 1.101, 1.102, 1.103, 1.104, 1.105, 1.106.

- 1.2 (a) Quantitative. This statement clearly involves a measurable distance.
(b) Qualitative. This is a value judgment. There is no numerical scale of measurement for artistic excellence.
(c) Qualitative. If the numerical values for the densities of ice and water were given, it would be a quantitative statement.
(d) Qualitative. Another value judgment.
(e) Qualitative. Even though numbers are involved, they are not the result of measurement.

- 1.4 (a) hypothesis (b) law (c) theory

- 1.9 Li, lithium; F, fluorine; P, phosphorus; Cu, copper; As, arsenic; Zn, zinc; Cl, chlorine; Pt, platinum; Mg, magnesium; U, uranium; Al, aluminum; Si, silicon; Ne, neon.

- 1.18 (a) Cs (b) Ge (c) Co (d) Sr

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