CLASSIFYING CHEMICAL REACTIONS

14. SO₃ + H₂O → H₂SO₄



Classify the following reactions as synthesis, decomposition, single replacement or double replacement. (AACCI = RC)

repr	acement.	ANSW	EKS/
1.	2KCIO ₃ → 2KCI + 3O ₂	Decon	nposition
2.	HCI + NaOH → NaCI + H ₂ O	Double	Replacement
з.	Mg + 2HCl → MgCl ₂ + H ₂	Single	Replacement.
4.	2H ₂ + O ₂ → 2H ₂ O	Synthesis	
5.	2AI + 3NIBr₂ → 2AIBr₃ + 3NI	Single	Replacement
6.	4AI + 3O₂ → 2AI₂O₃	Synth	ne Sis
7.	2NaCl → 2Na + Cl ₂	Decom	position
8.	$CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$	Single	Replacement
9.	AgNO ₃ + KCI → AgCI + KNO ₃	Double	Replacement
10.	N ₂ + 3H ₂ → 2NH ₃	Synth	nesis
11.	2H ₂ O ₂ → 2H ₂ O + O ₂	Decomp	position
12.	$(NH_4)_2SO_4 + Ba(NO_3)_2 \rightarrow BaSO_4$	+ 2NH4NO3 DO	uble Replacement
13.	Mgl ₂ + Br ₂ → MgBr ₂ + l ₂	Single	Replacement

Synthesis

General Chemistry Massing Activities Lab Answers

Thomas Griffiths

General Chemistry Massing Activities Lab Answers:

Reviewing General Chemistry Massing Activities Lab Answers: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "General Chemistry Massing Activities Lab Answers," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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